

Fysikken bag hverdagens materialer.



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Oversigt

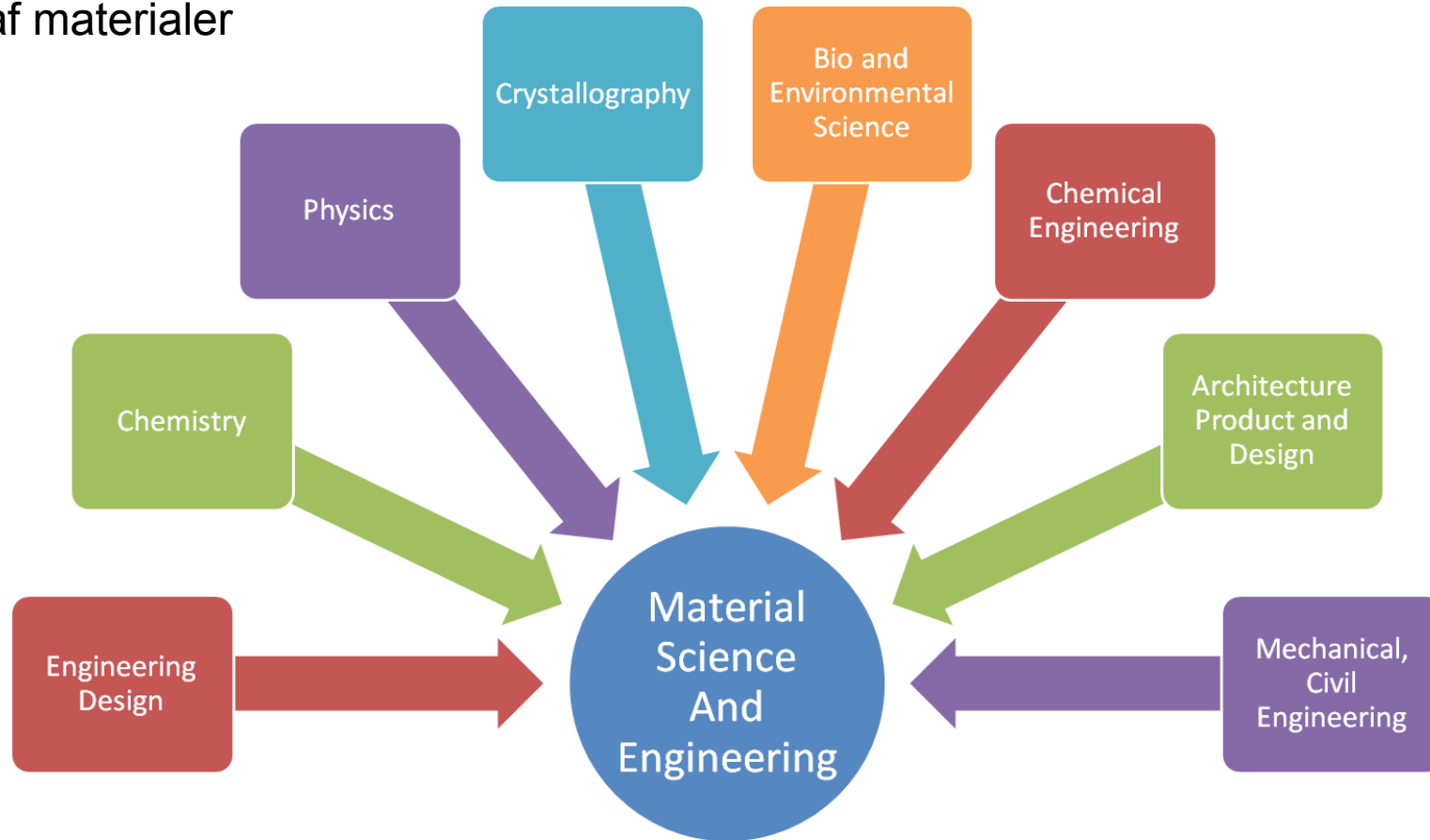
- Intro hverdagens materialer (temperatur, brownsk bevægelse)
- Surfaktanter (hydrofob/hydrofil, overflade spænding, selv-samling)
- Polymerer (elasticitet, gel overgang)
- Colloider (stabilitet emulsioner)
- Væsker/glas (viskositet, dynamik)
- Biologiske materialer

Materialevidenskab

Materialevidenskab er studiet af

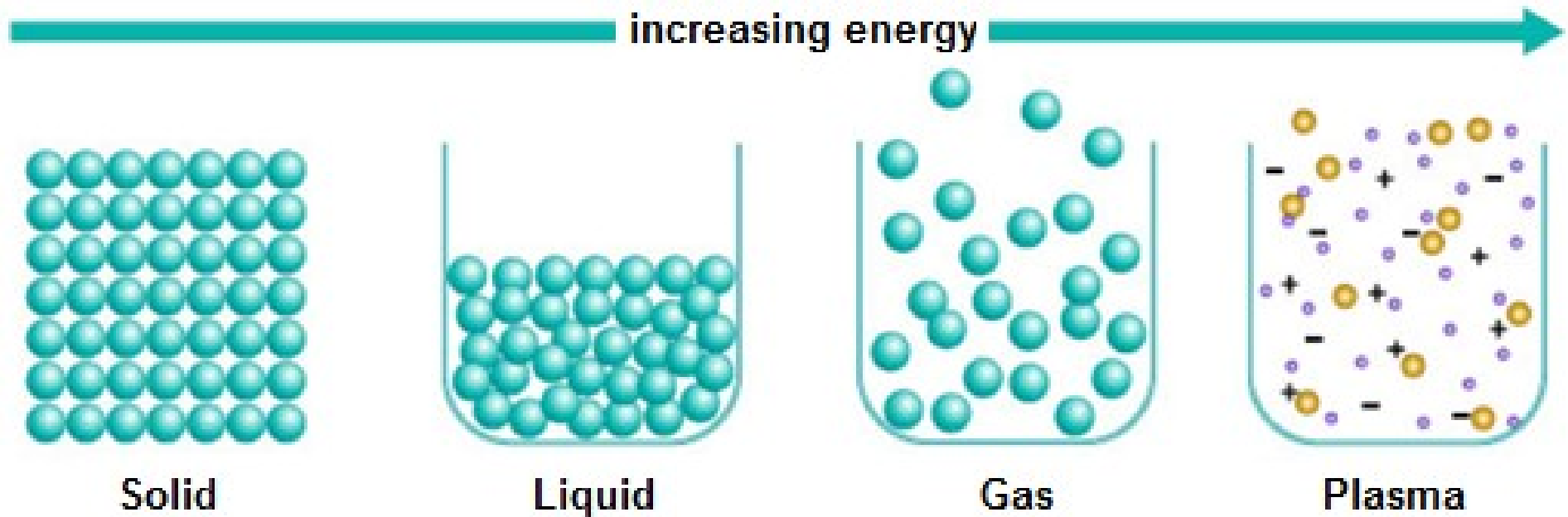
- Struktur og vekselvirkning mellem molekulære byggesten
- Forståelse af materialers egenskaber
- Design af materialer

Materialer dækker her både over væsker og faste stoffer.

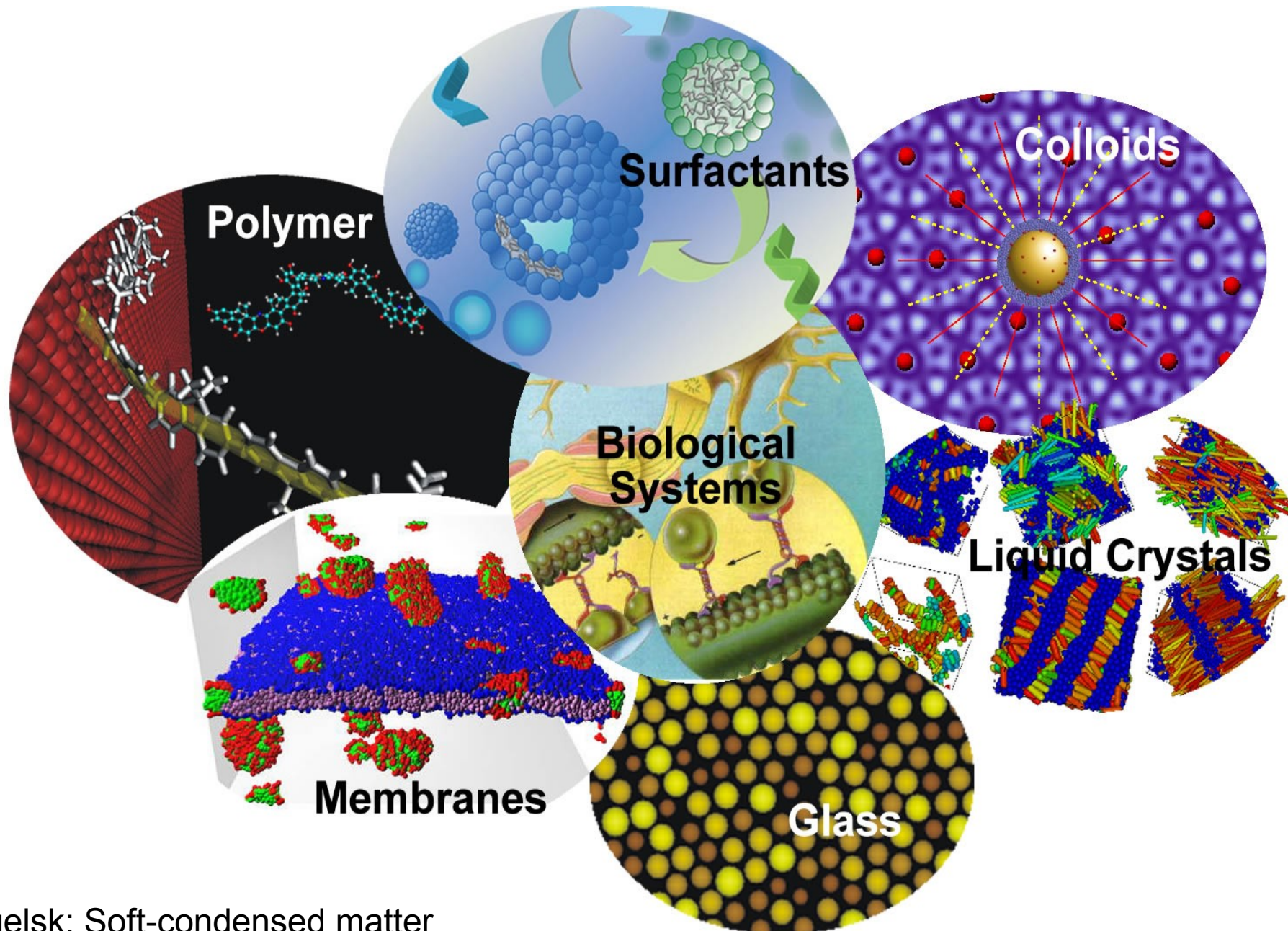


Og mange mange flere...

Tilstandsformer

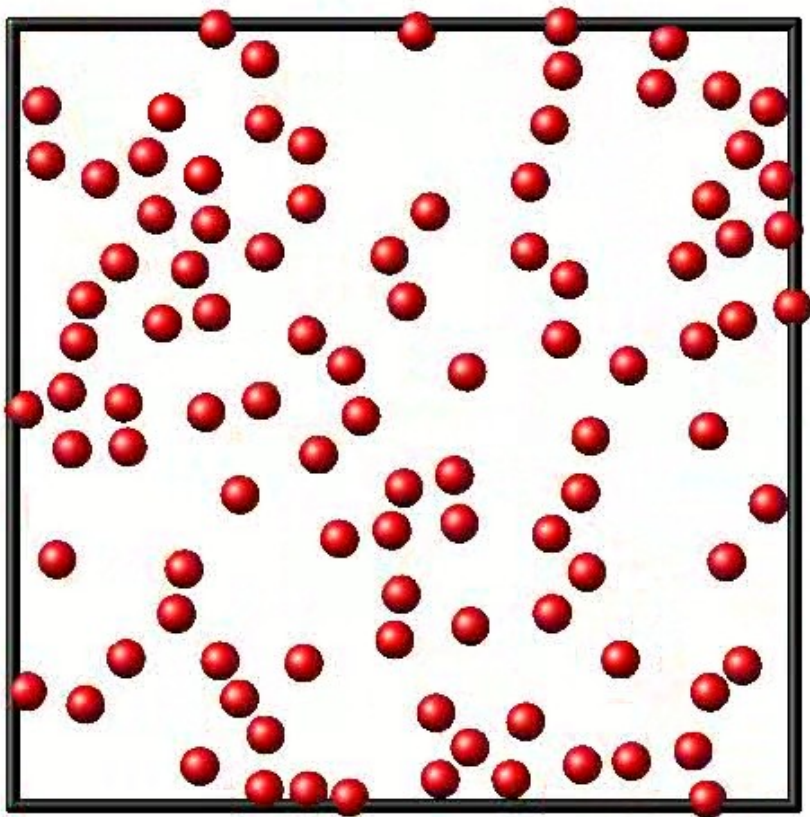


Hverdagsmaterialer

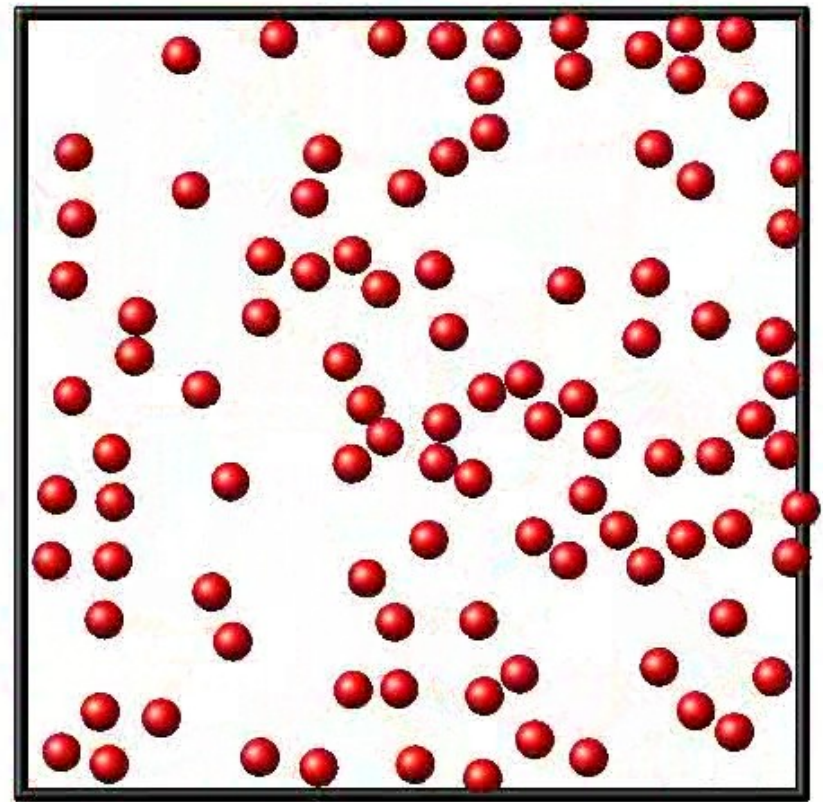


Engelsk: Soft-condensed matter

Temperatur



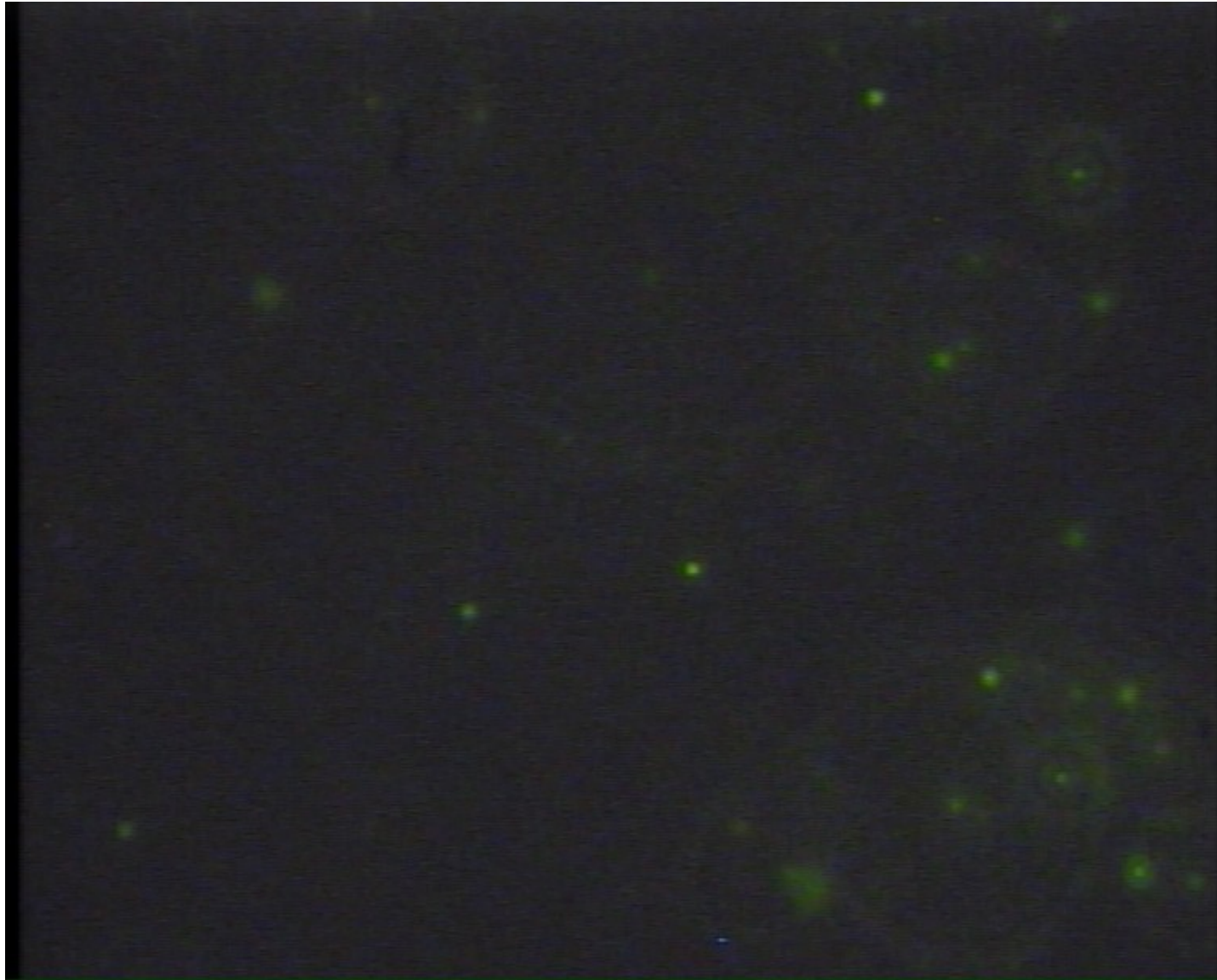
T lav



T høj

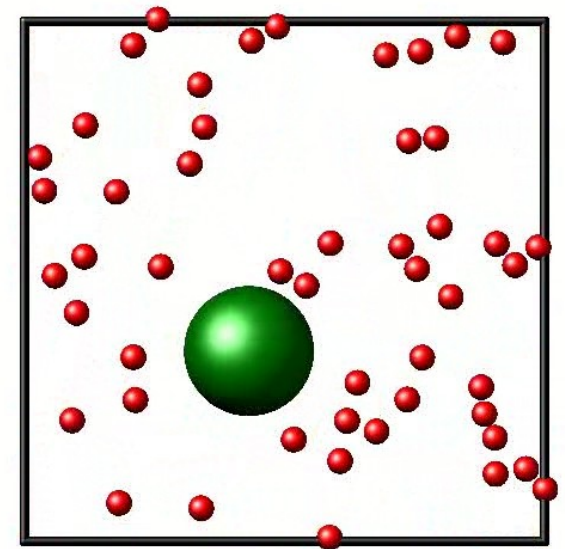
$$\langle E_{kin} \rangle = \frac{1}{2} m \langle v^2 \rangle = \frac{1}{2} k_B T$$

Brownsk bevægelse



Eksperiment

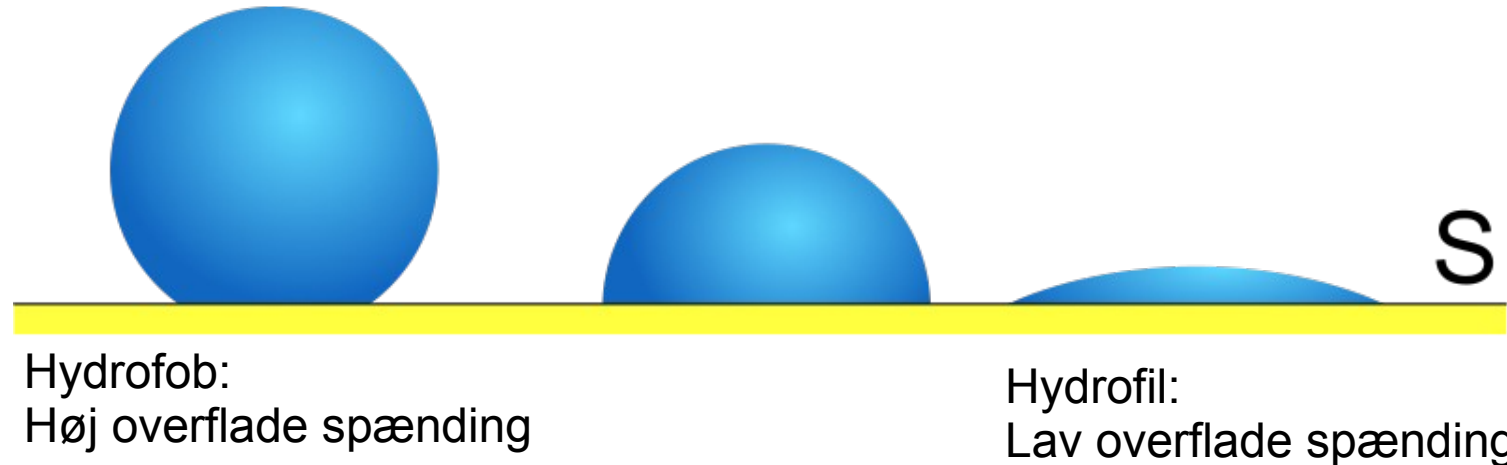
Simulation



Hydrofob/hydrofil

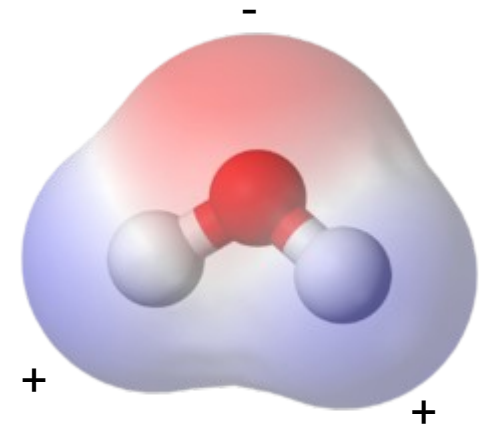
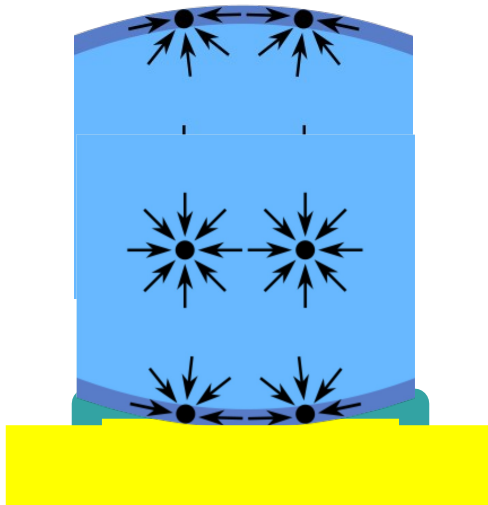
Energibidrag
fra kontraktfladen

$$E = \gamma A$$

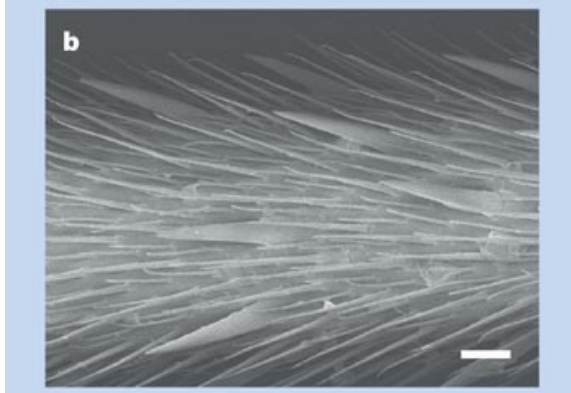
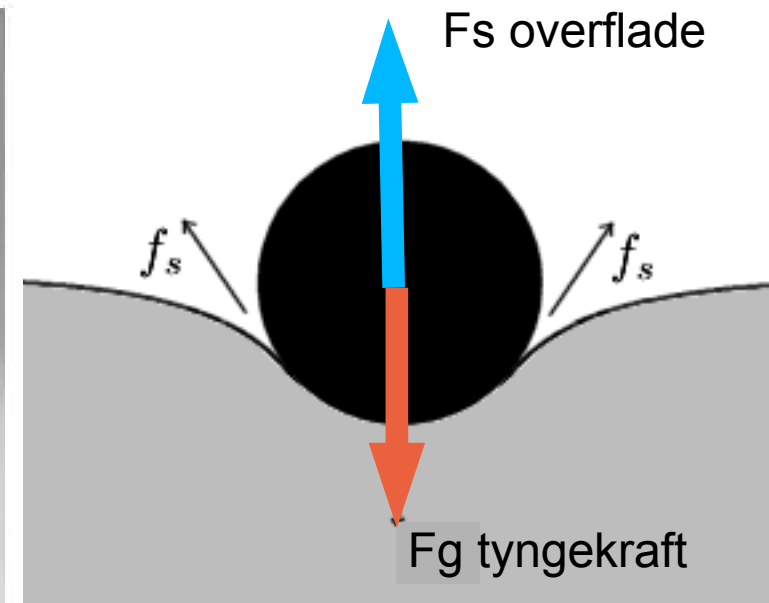
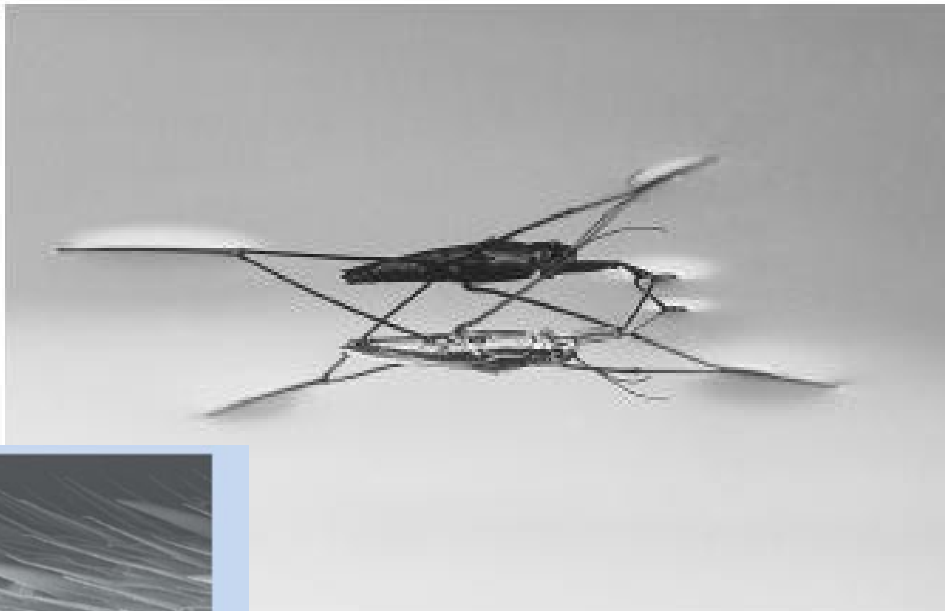


Færre molekulære velsekvirkninger ved
overflader end for molekyler inde i væsken.

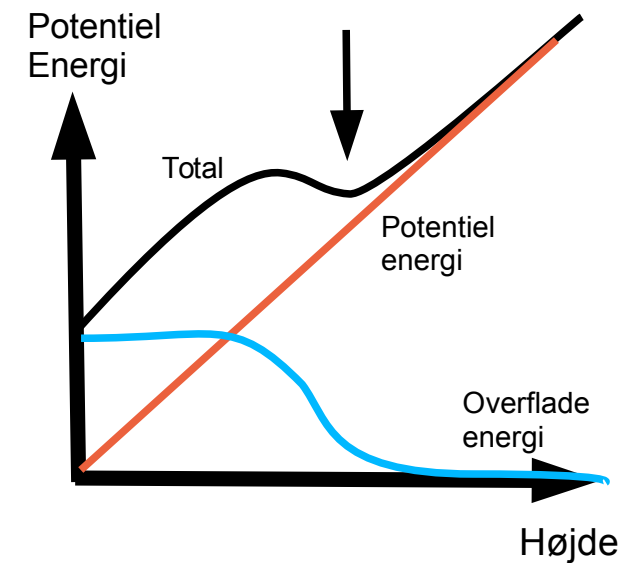
Derfor har molekylerne typisk højere
energi ved grænsefladen.



Gå på vand!



Øg overflade
arealet!



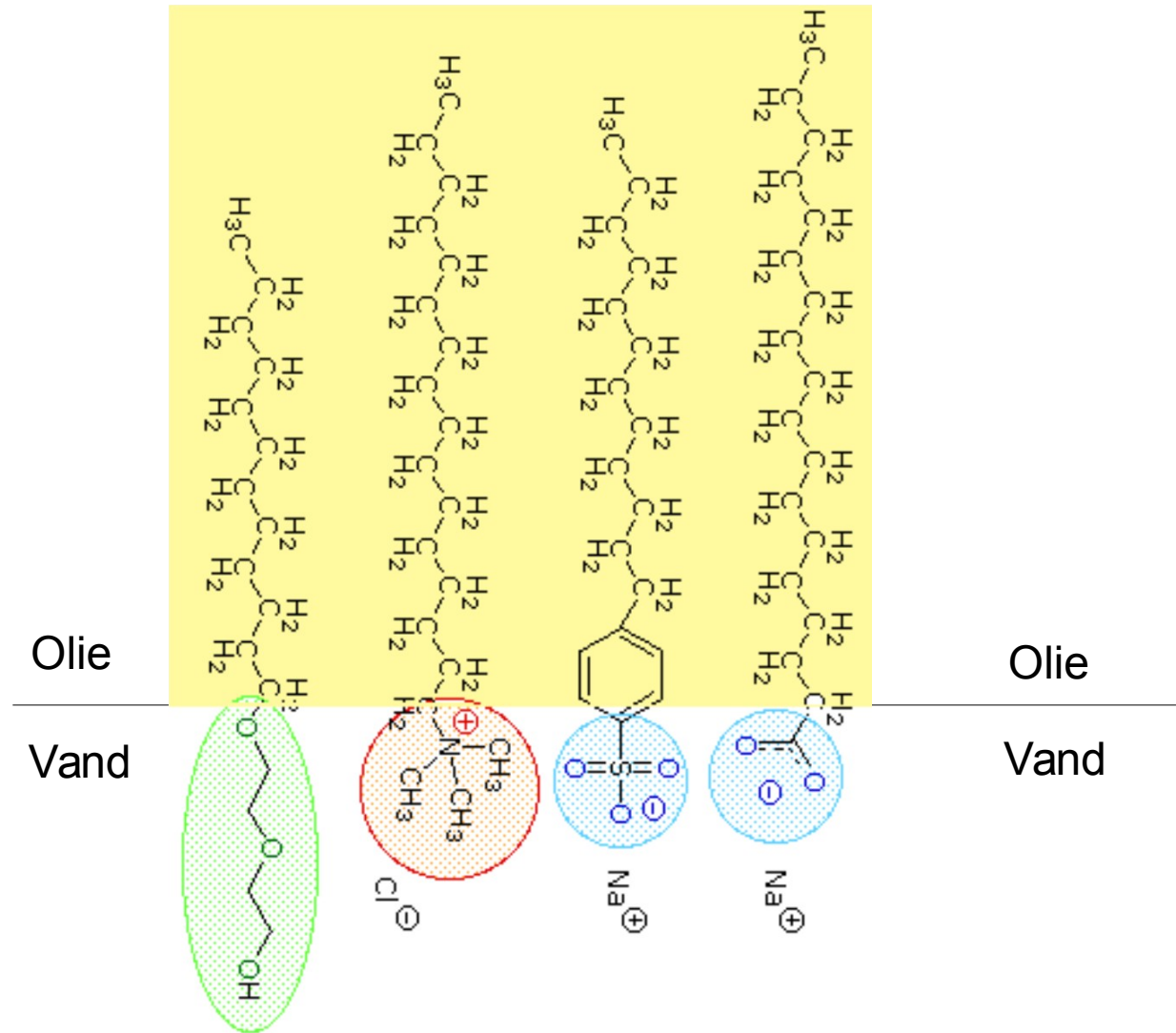
Superhydrofob



Kilde youtube: The SECOND Official Ultra-Ever Dry Video - Superhydrophobic

Surfaktanter

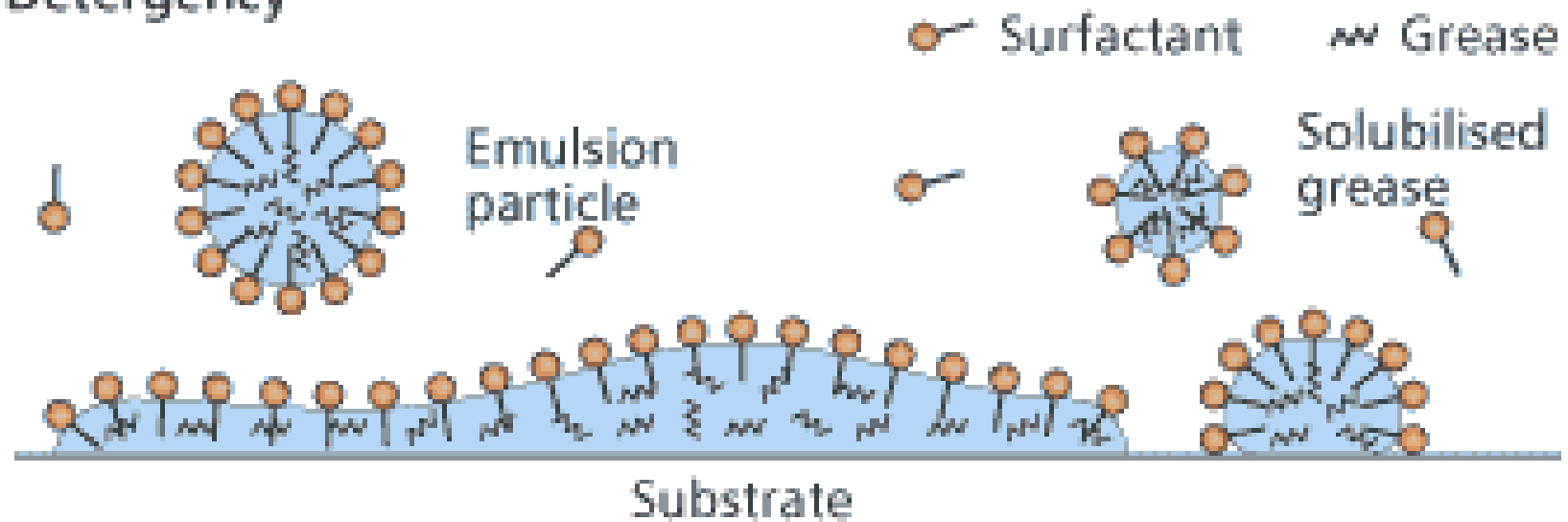
Surfaktant = SURface Aktive AgeNTer, dvs. Molekyler, der ændre egenskaber af overflader.



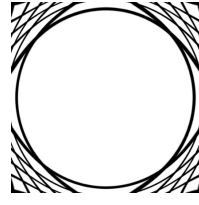
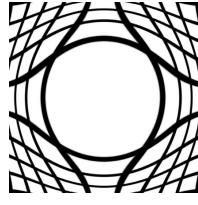
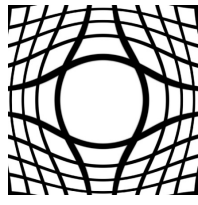
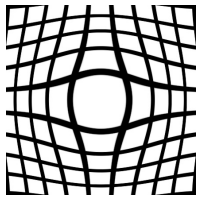
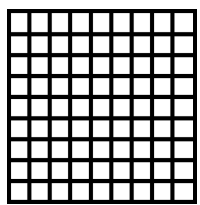
Produkter: Sæbe, Shampoo, vaskemiddel, opvaskemiddel, ...

Anvendelse: Opvask

Detergency



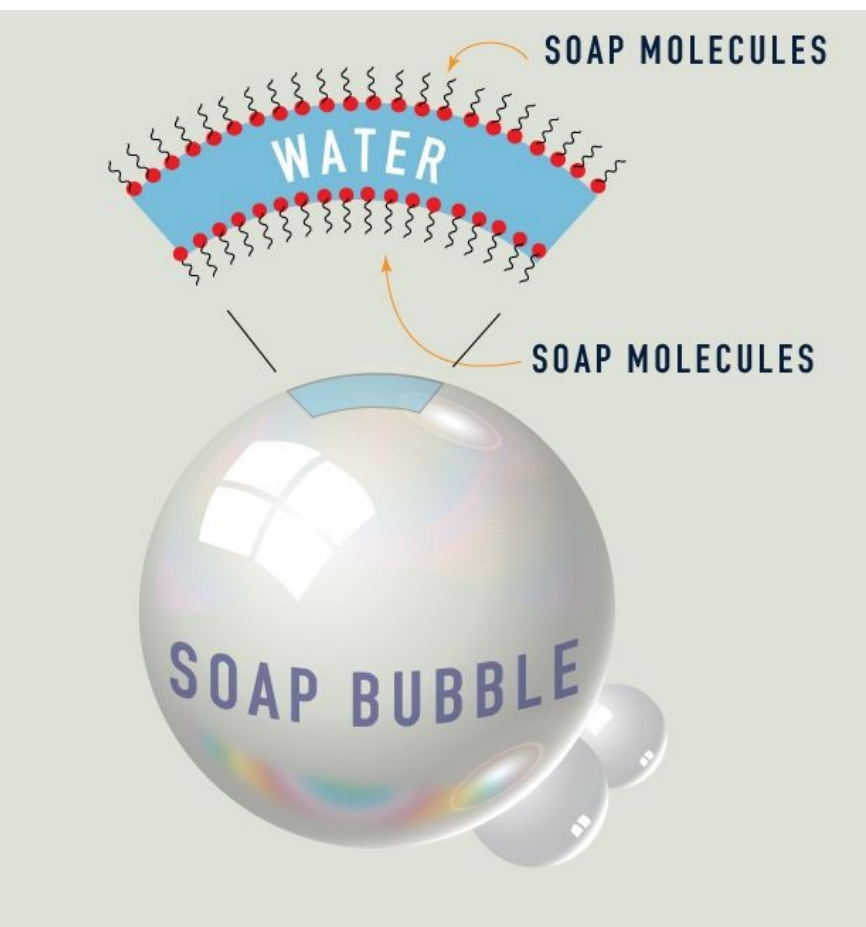
Surfaktanter sænker overfladespænding



Kilde youtube: The marangoni effekt: how to make soap boats

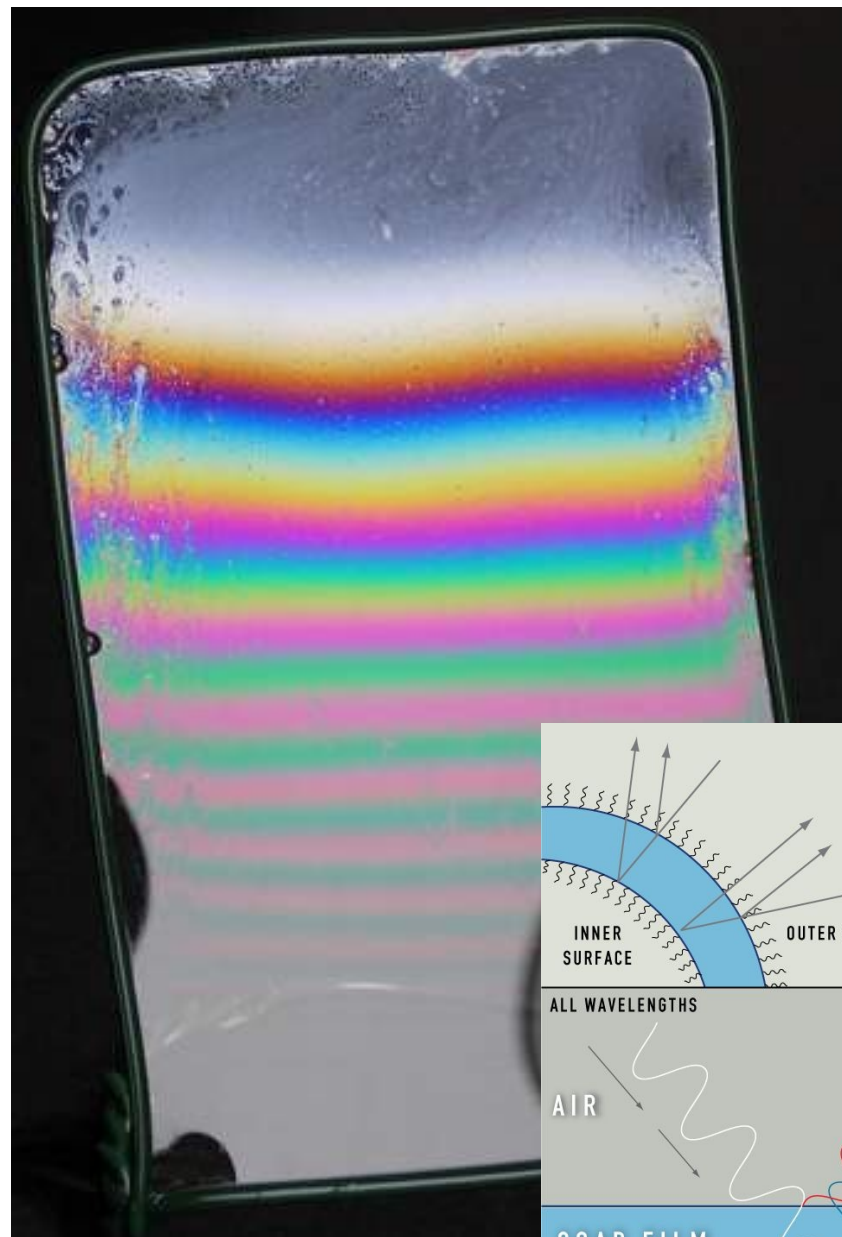
Opskrift:
Vand
Opvaskemiddel

Sæbebobler

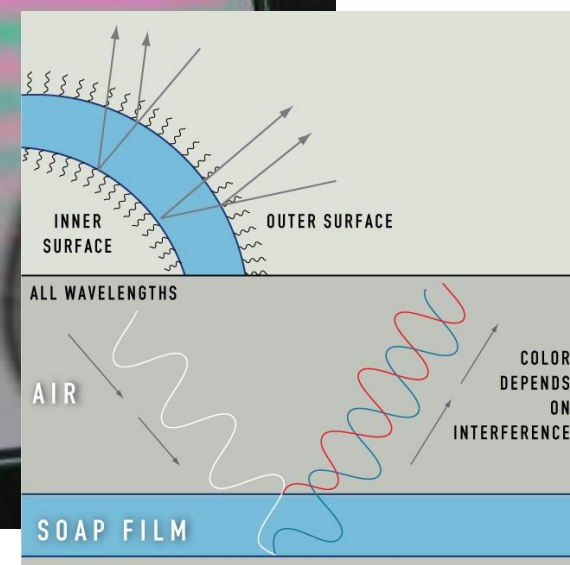


Opskrift:

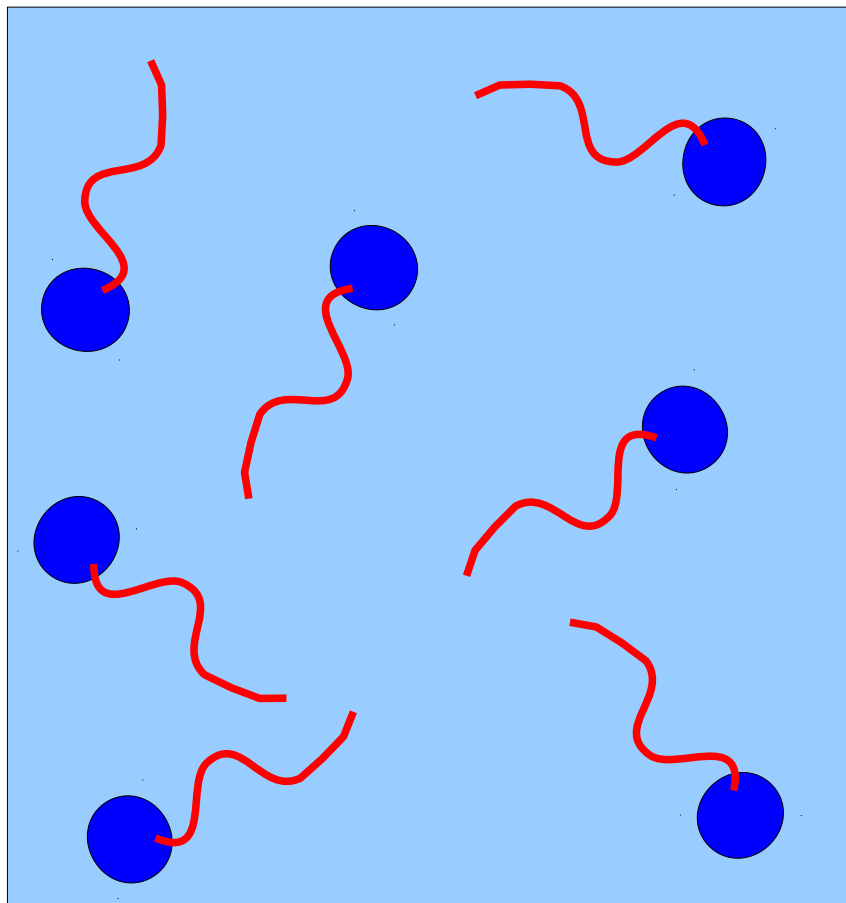
Vand
Opvaskemiddel
Glycerin



Interferens



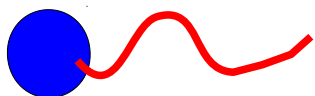
Selv-bygning



Surfaktanter i opløsning.

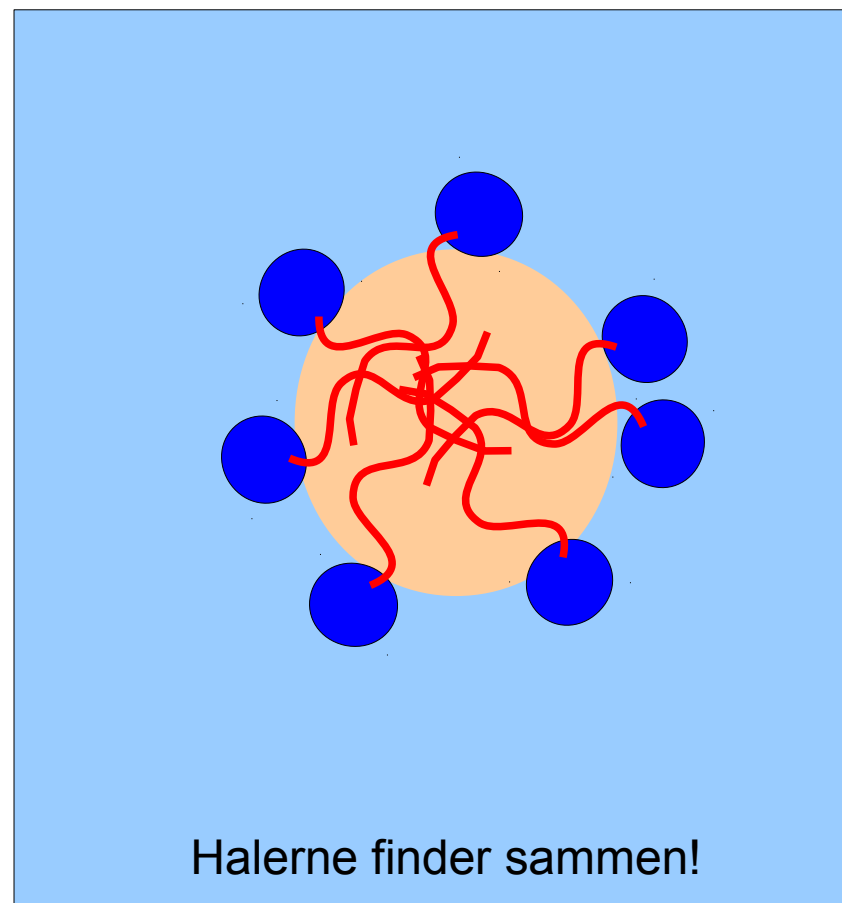
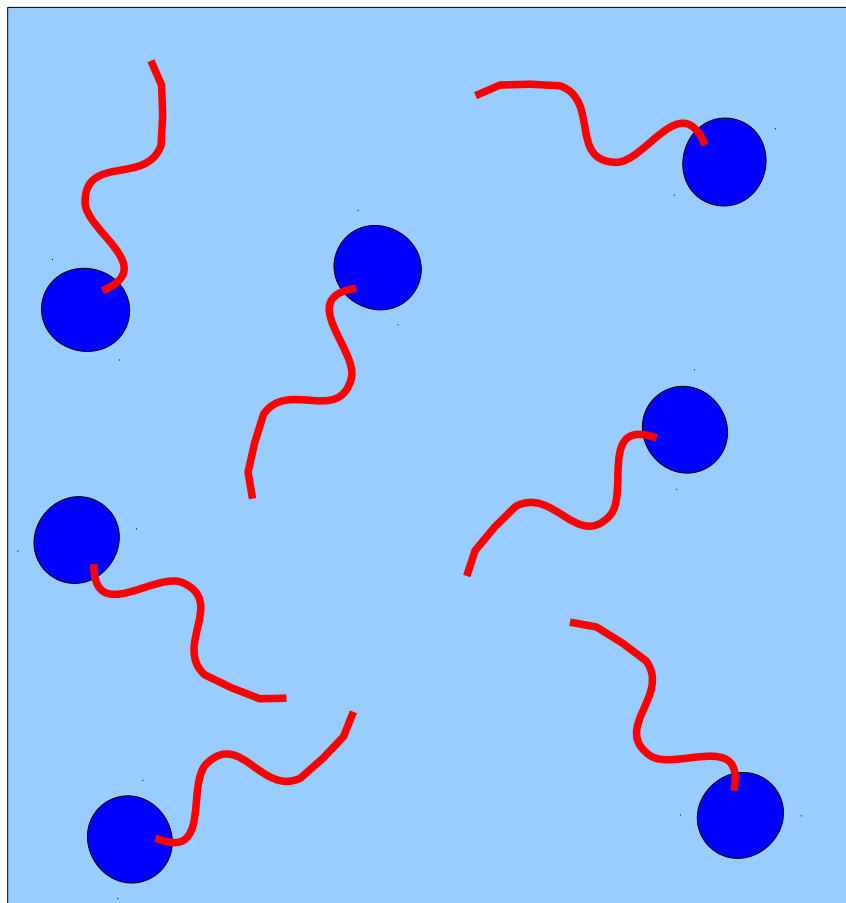
Halerne er stadig i kontakt med vand,
Det kan de ikke lide.

Hvad kan surfaktanterne gøre ved
det?



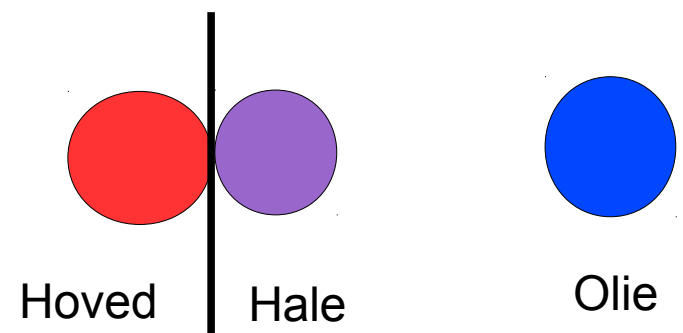
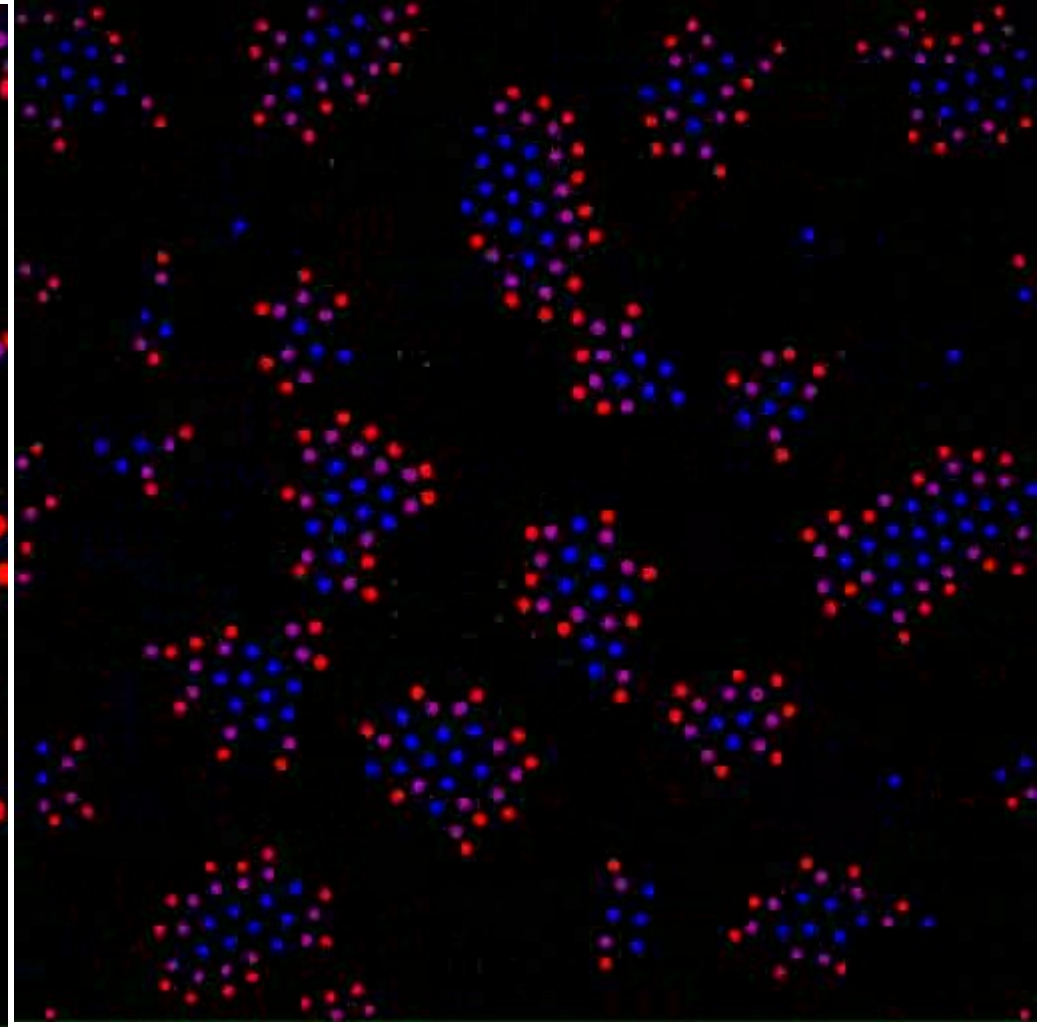
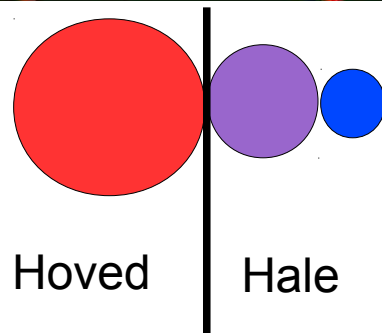
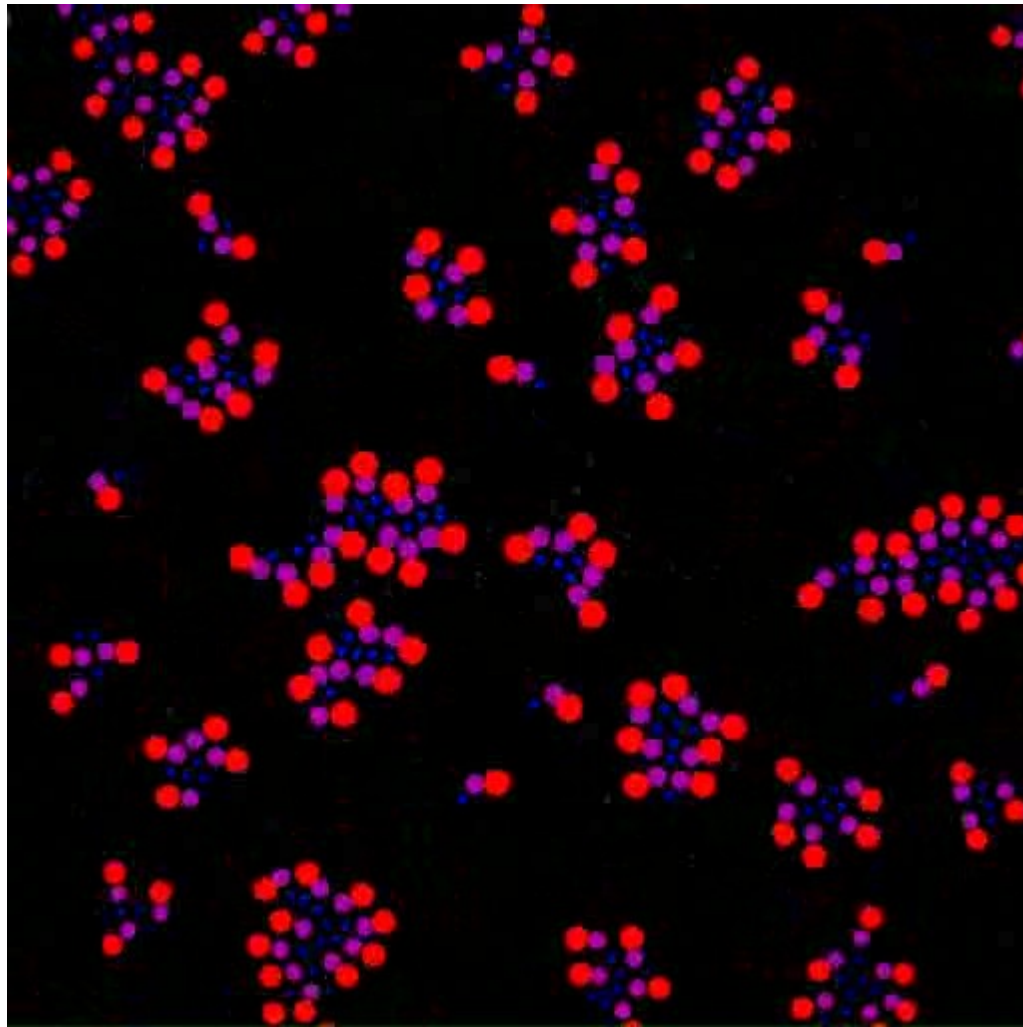
hydrofil hydrofob

Selv-bygning



Molekylære legoklodser! Bygger selv struktur på det molekylære niveau.

Simulation af surfaktanter



Self-byggede strukturer

$$\frac{v}{a|} < \frac{1}{3}$$



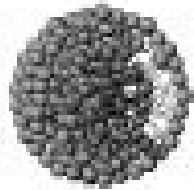
$$\frac{1}{3} < \frac{v}{a|} < \frac{1}{2}$$



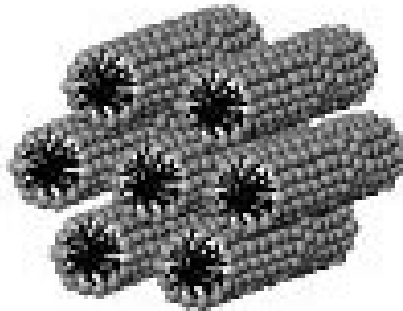
$$\frac{v}{a|} \sim 1$$



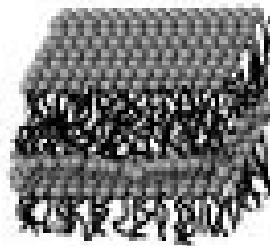
$$\frac{v}{a|} > 1$$



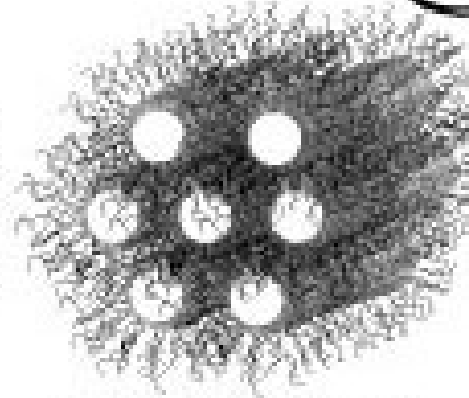
Micelle (L_1)



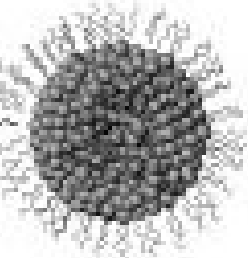
Hexagonal (H_1)



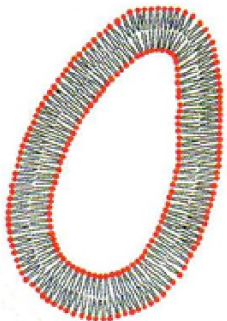
Lamellar (L_α)



Inverse Hexagonal (H_2)



Inverse Micelle (L_2)



Vesikel

Selv-bygning / Emergens

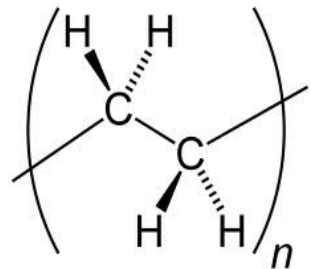
Hvor mange X skal man have for at have en Y?



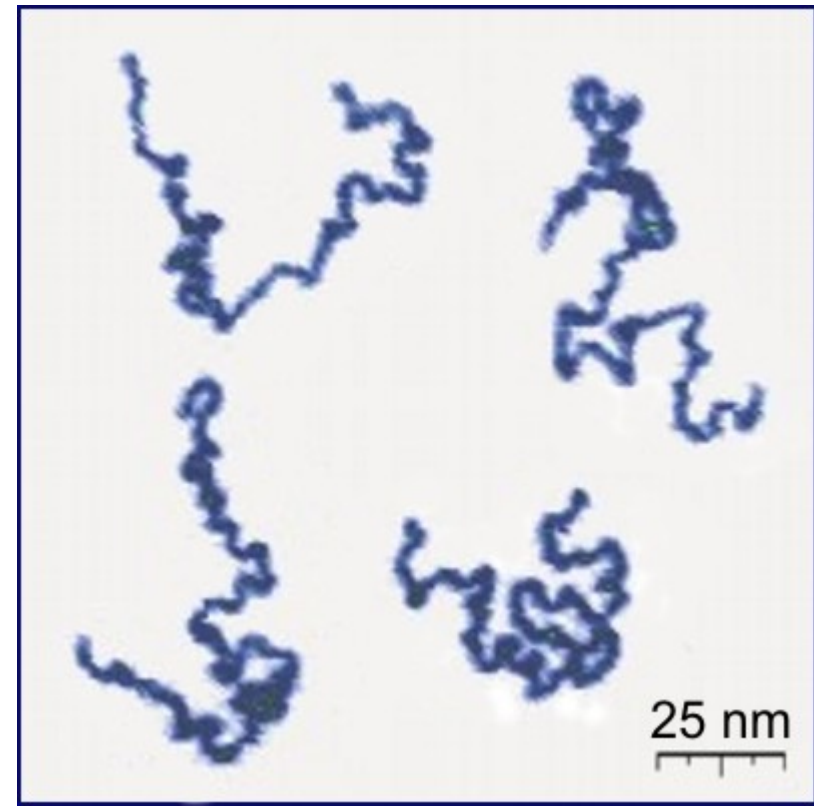
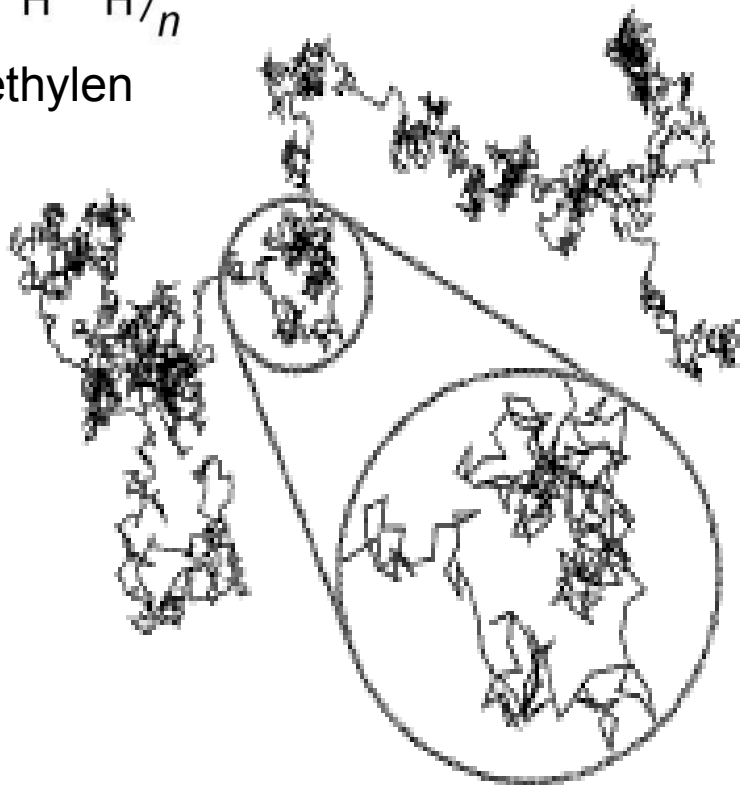
Helheden er mere end summen af komponenterne!

Polymere

Poly = mange,
meros = dele

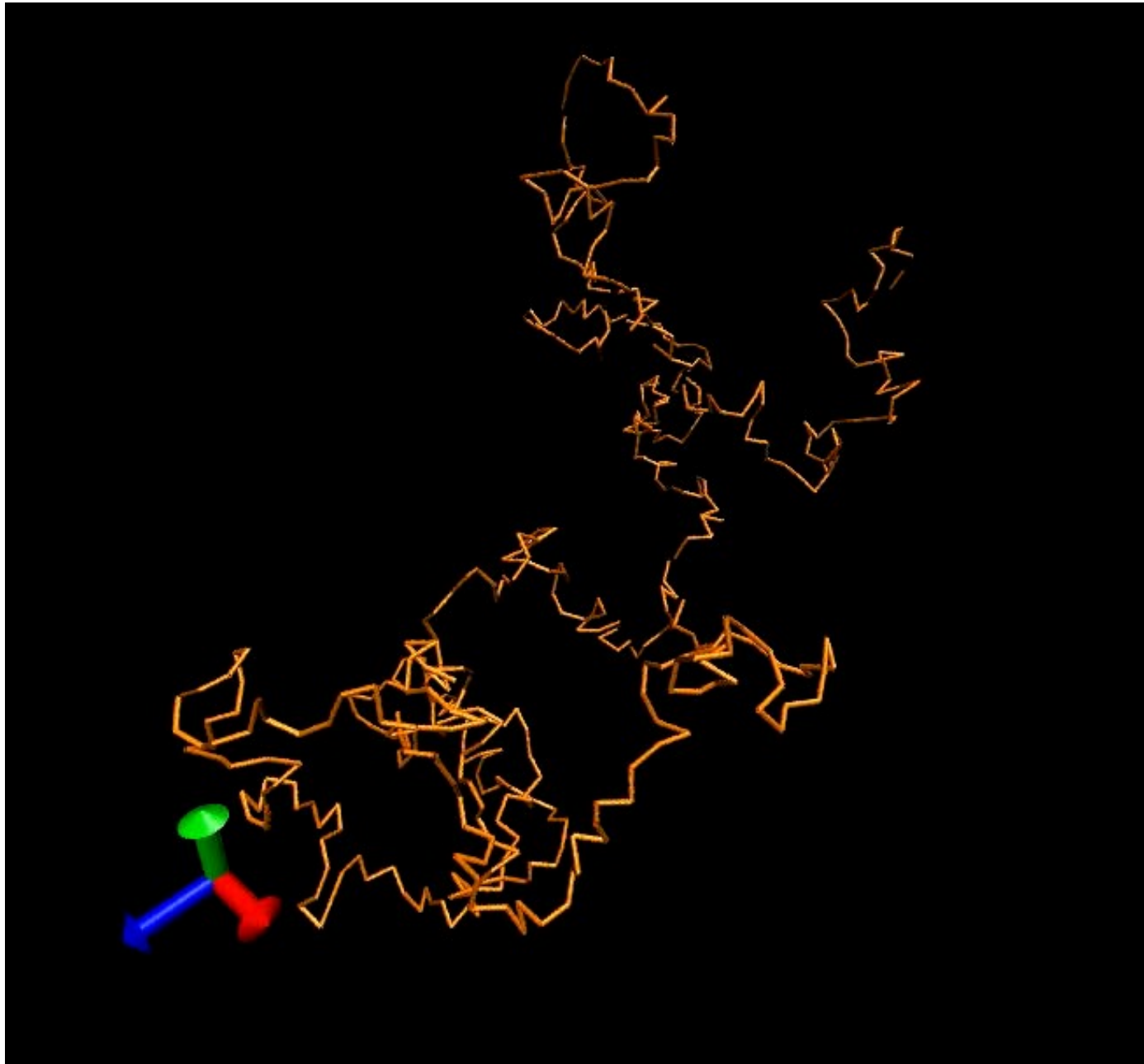


polyethylen



Atomic-Force mikroskop billed
af polymere på en overflade.
Bredde ~ 0.4nm, længde 204nm
(Kilde: Wikipedia)

Brownsk bevægelse

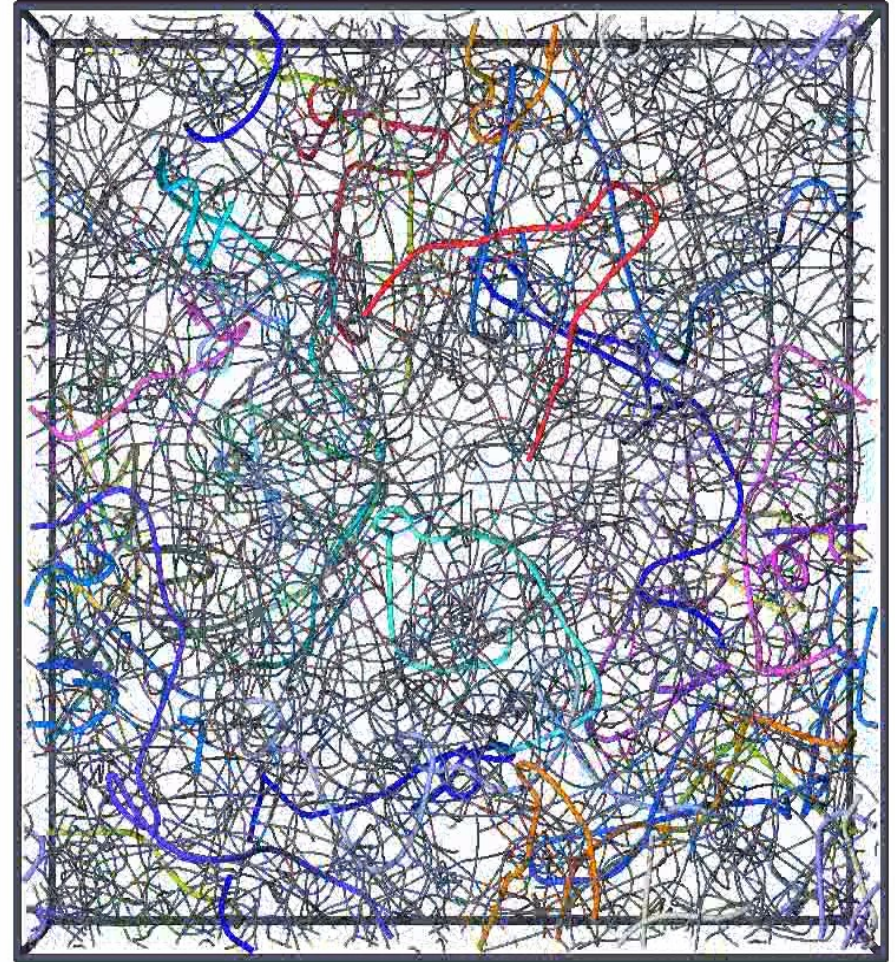
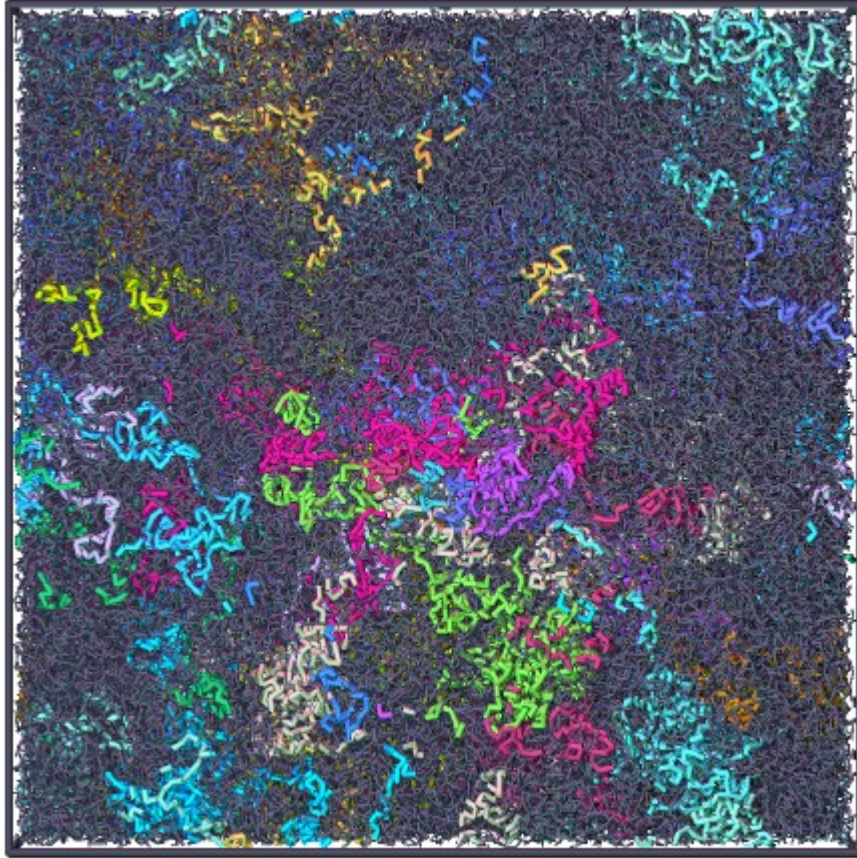


Anvendelser

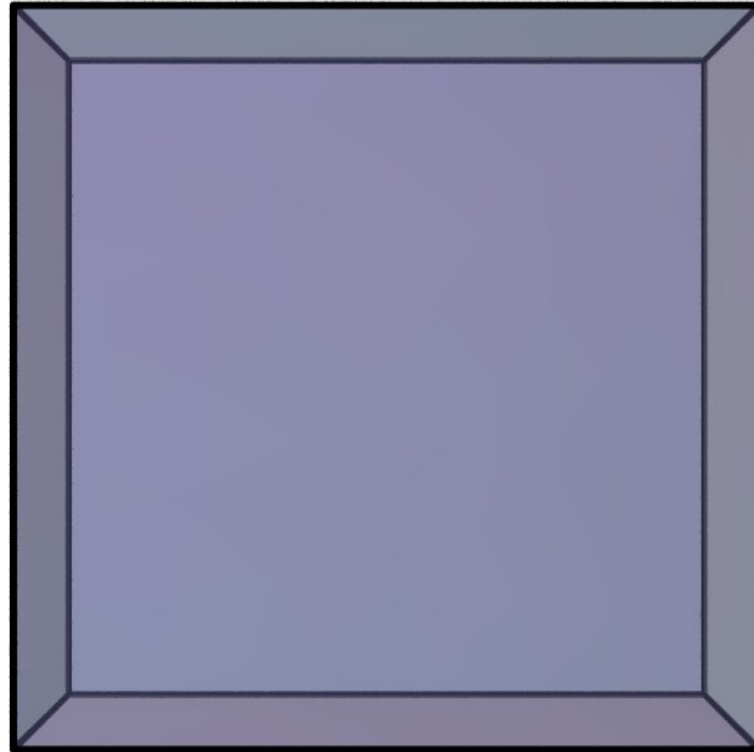


For at nævne nogle få...

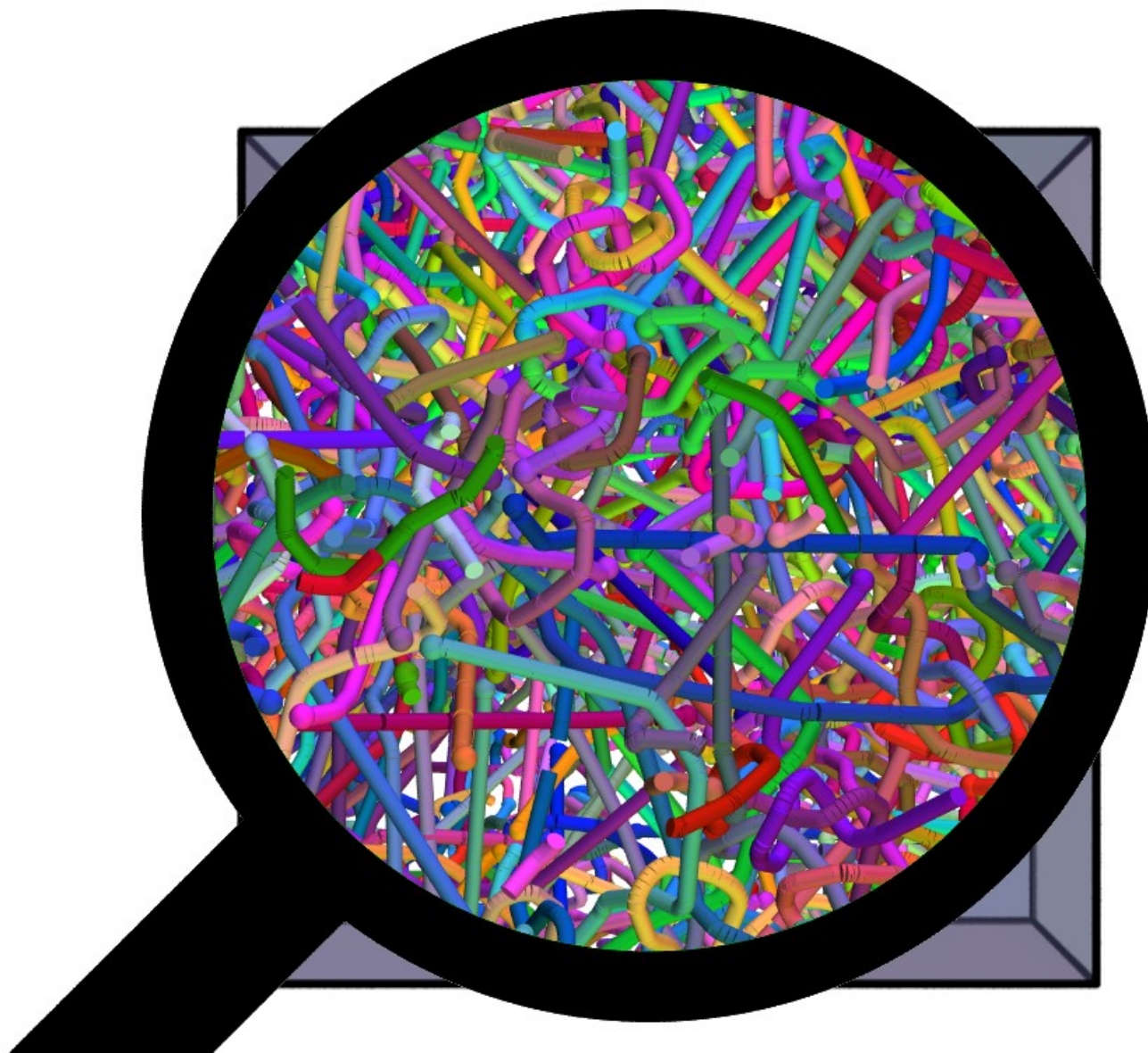
Gummi: Høj tæthed



Gummi

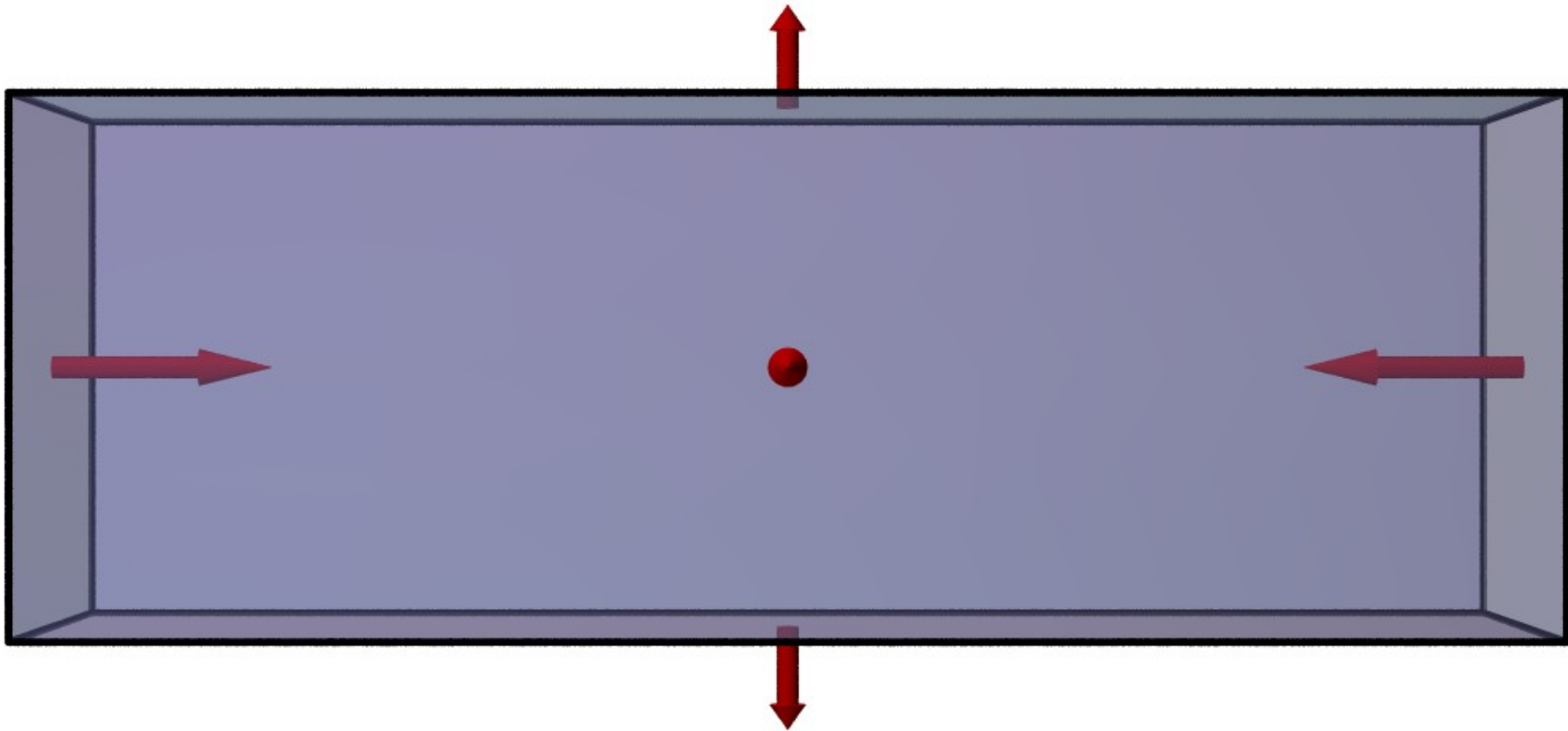


Gummi





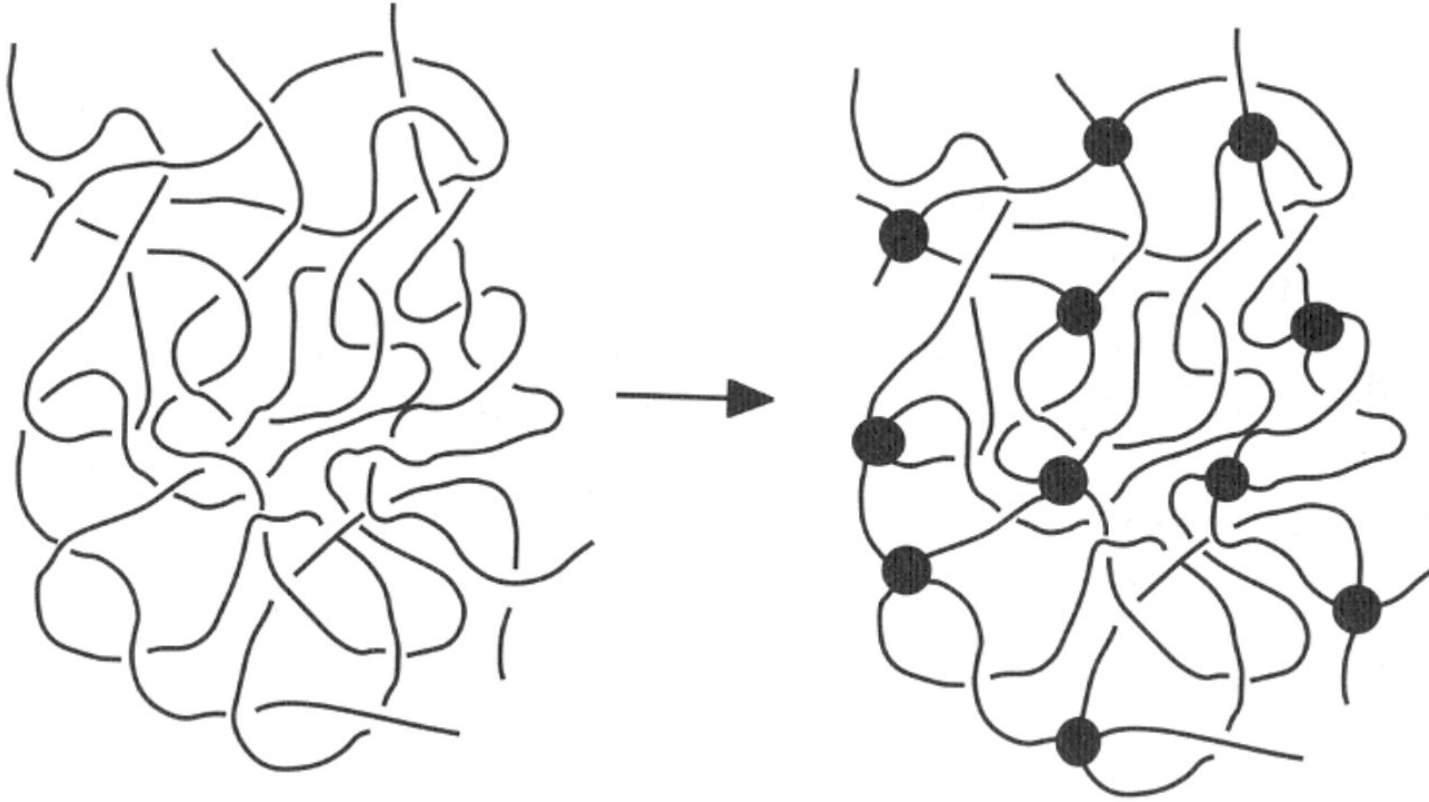
Gummi



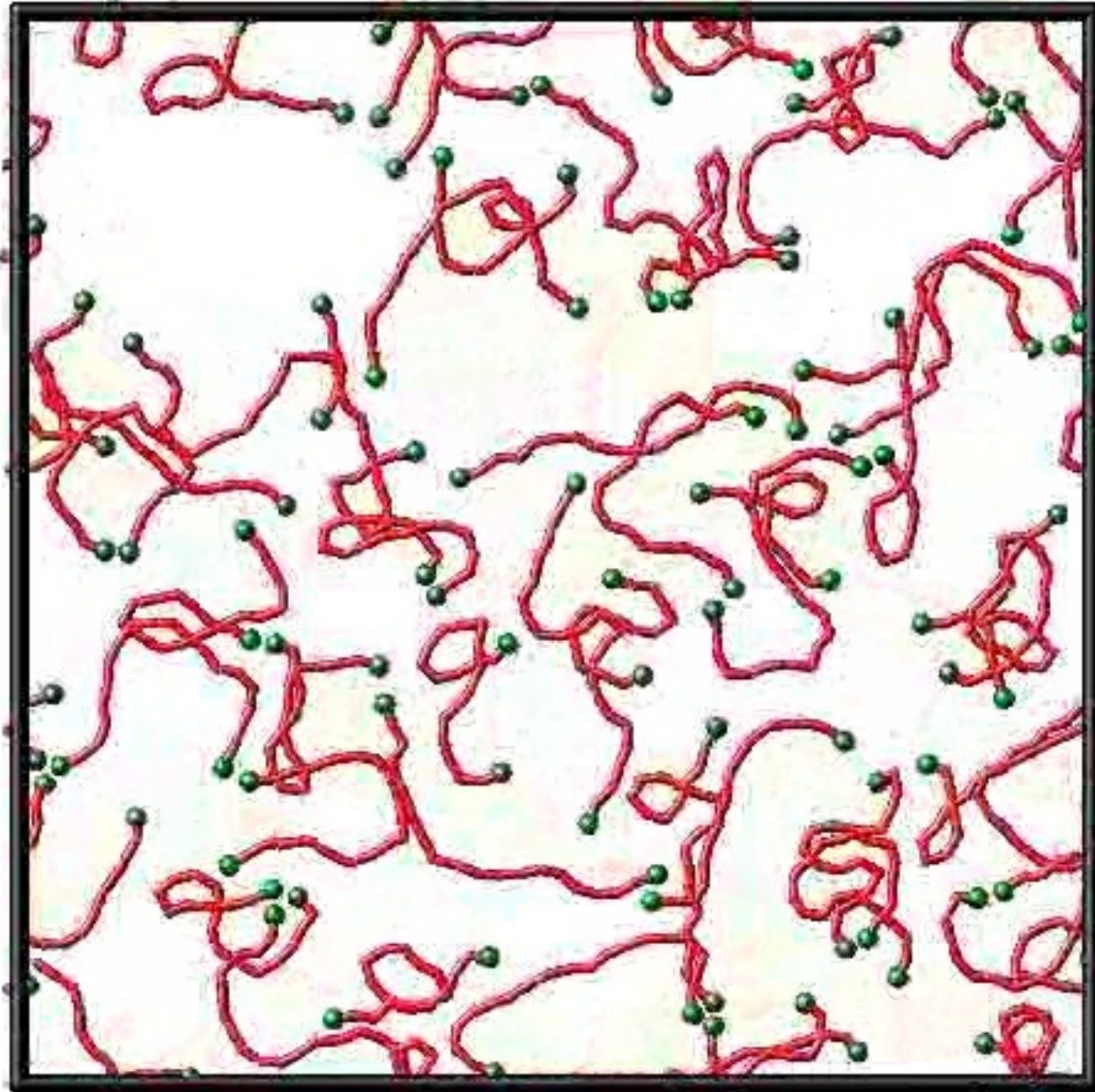
Gummi kan trækkes x10 fordi molekylerne er fleksible! Shape-memory material.

Geler

Lavere tæthed af polymerer. Krydslænkes



Gelation



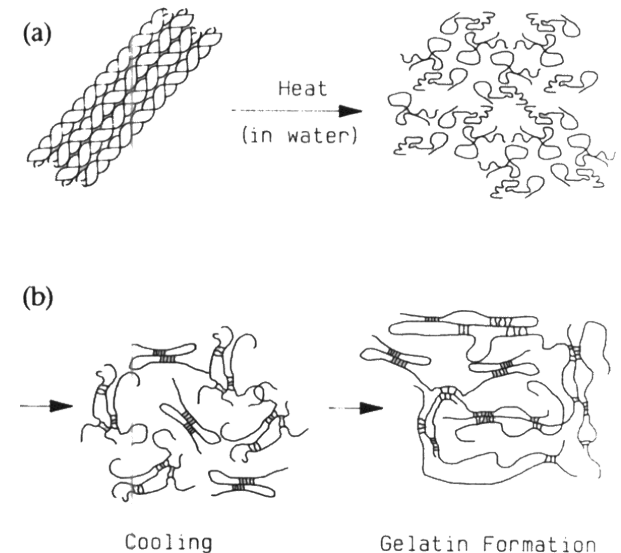
Polyakryl



Superabsorberende materialer, nyttig i f.eks. bleer!

Vingummi opskrift:

Bland Gelatine+smag +varmt vand
Hæld væsken I forme og køl ned.



Viskoelastisk materiale



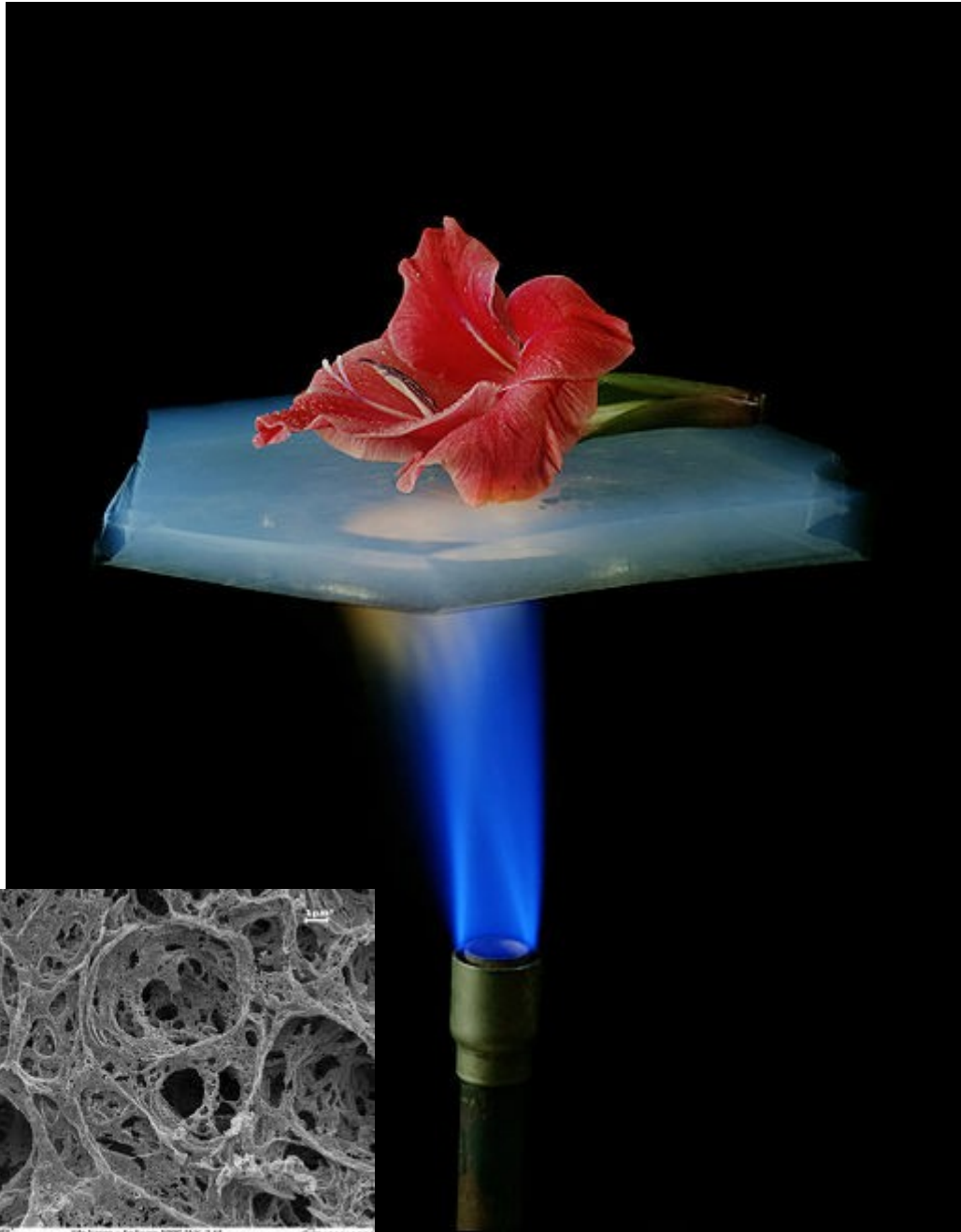
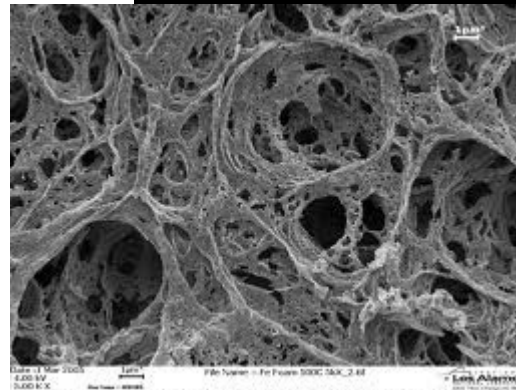
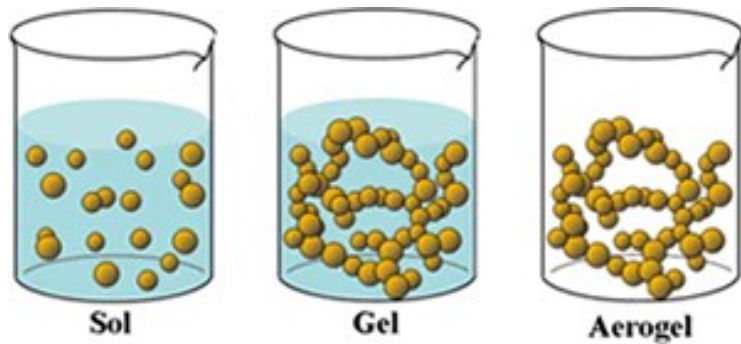
Colloider

Opløsning af X i medium Y, og størrelsen af X typisk er 1mn-1um.	Gas	Væske	Fast
Gas	-	Skum: Piskefløde, baberskum	Aerogel, flamico, Pimpsten
Væske	Væske aerosol: Tåge, hairspay	Emulsioner: Mælk, Mayonaise	Gel: Agar, vingummi, budding.
Fast	Fast aerosol: røg, is i skyer	Sol: Blæk, blod, Maling	Fløde is

Aerogel

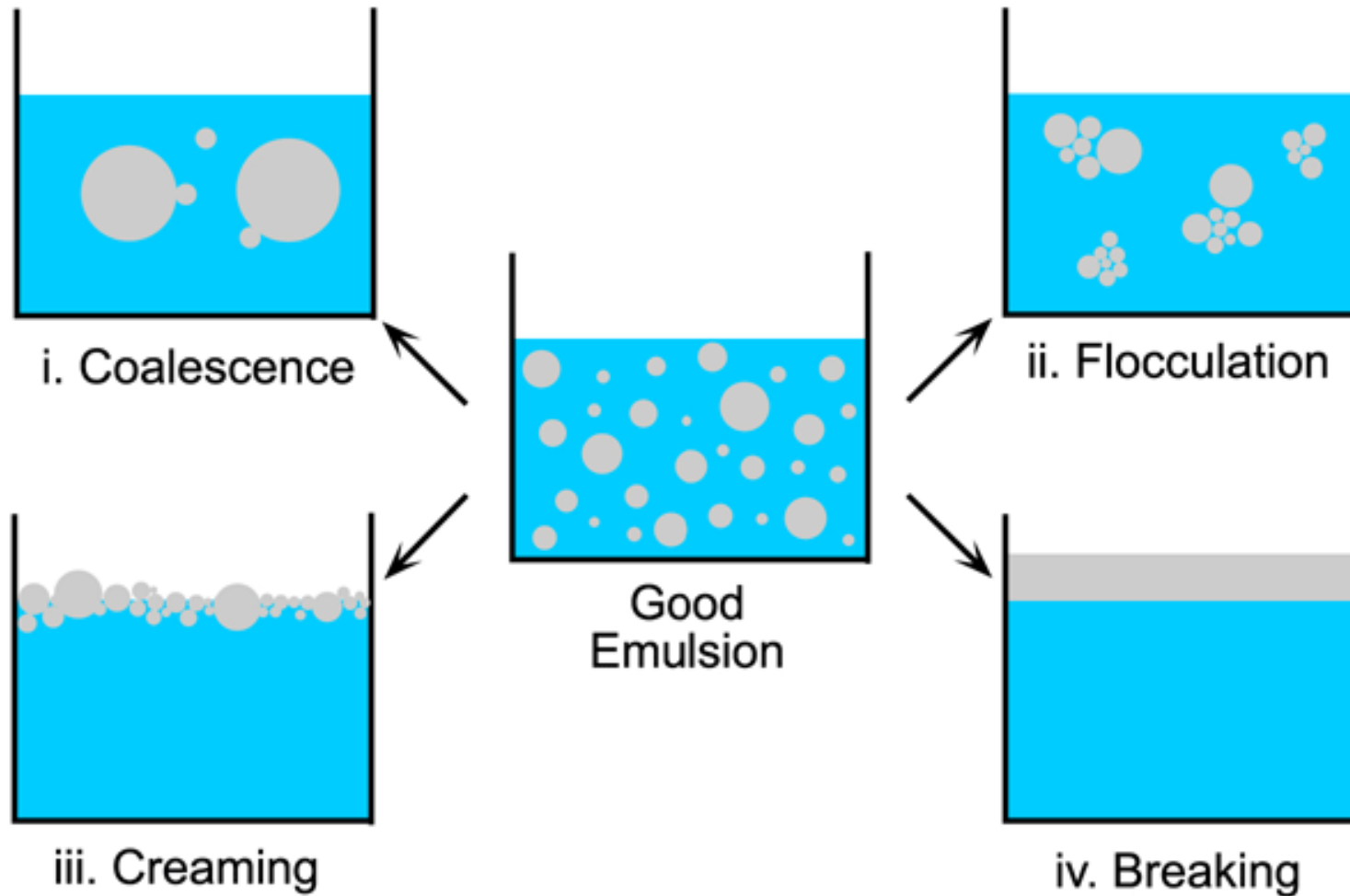


Ultralet materiale 99.98% luft
Tæthed 2 g/m^3 (luft vejer $1,2 \text{ g/m}^3$)
Polymer netværk af SiO_2

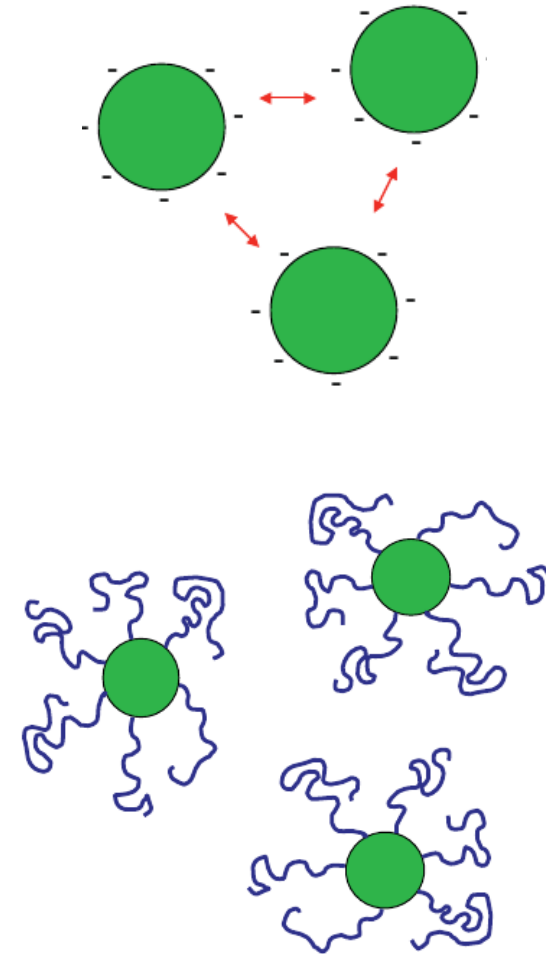


Emulsion stabilitet

Emulsion: Væske opløst i væske.



Stabilisering:



Emulsioner

Mayonaisse:

25% Vand f.eks. citron saft / eddike

75% Olie f.eks oliven olie

Surfaktant i form af æggeblomme.

Opskrift:

Start med at bland æg + eddike.

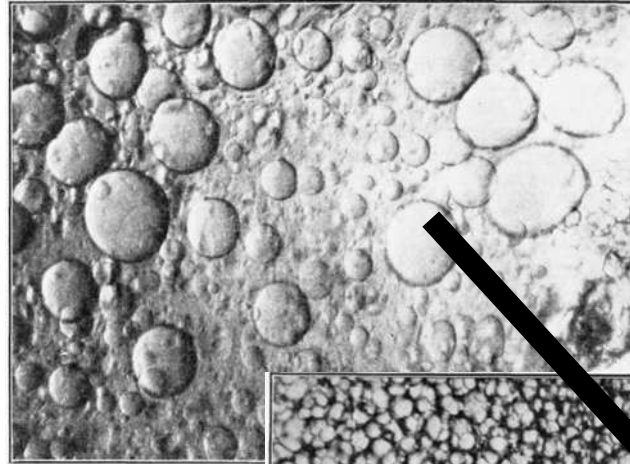
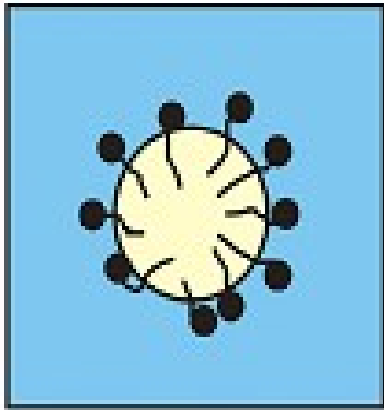
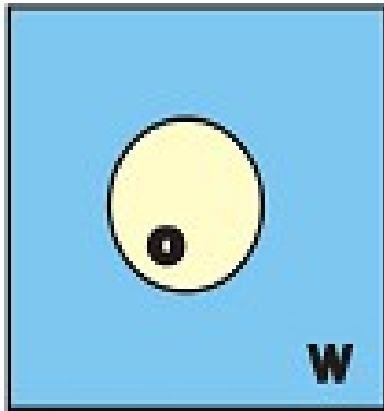
Tilsæt olie gradvist under omrøring.

Salt+peber til sidst.

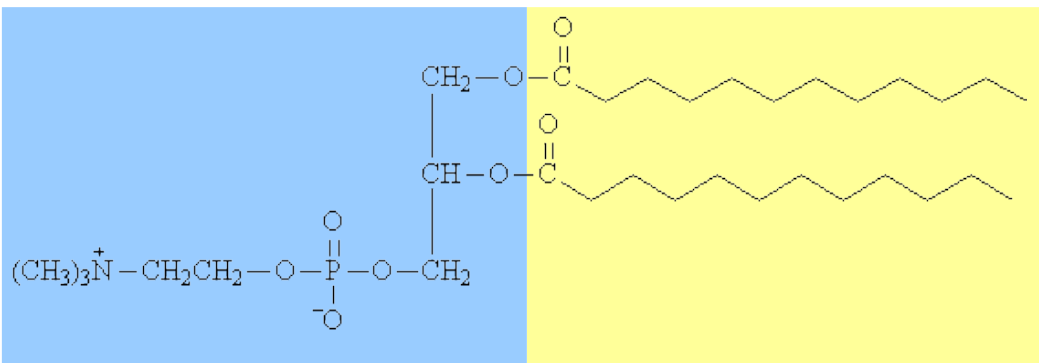
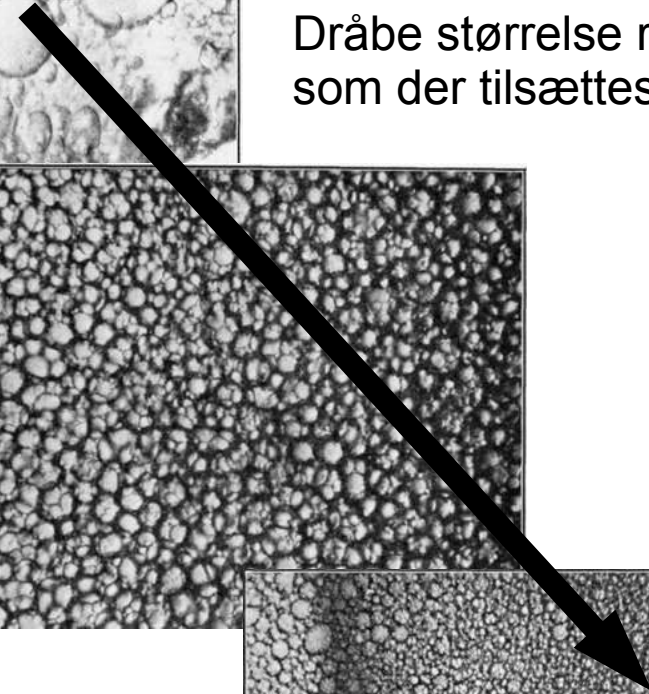
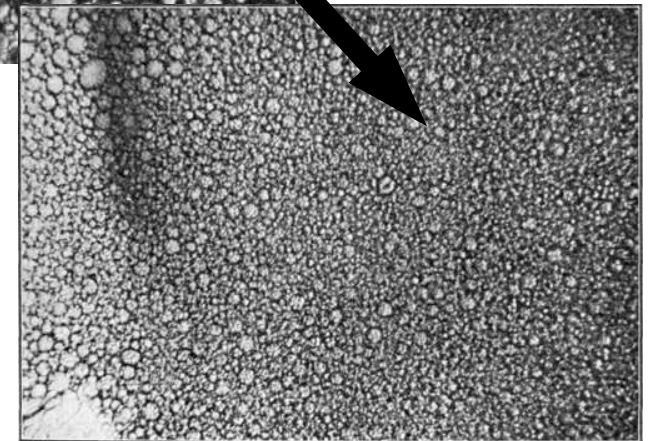
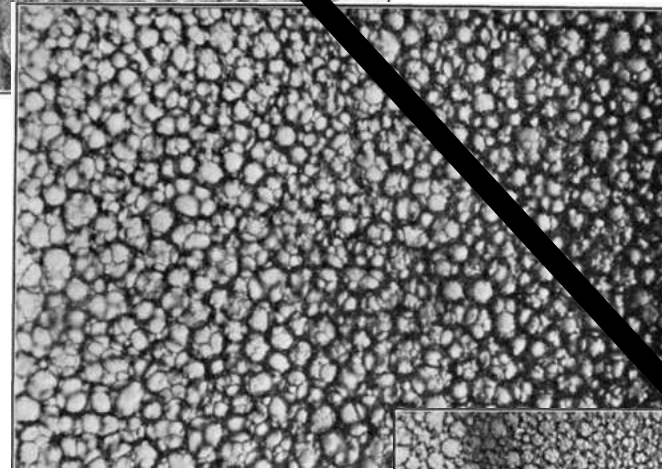
Vigtig alle ingredienser skal have samme temperatur.



Mayonaisse



Dråbe størrelse reduceres som der tilsættes mere olie.

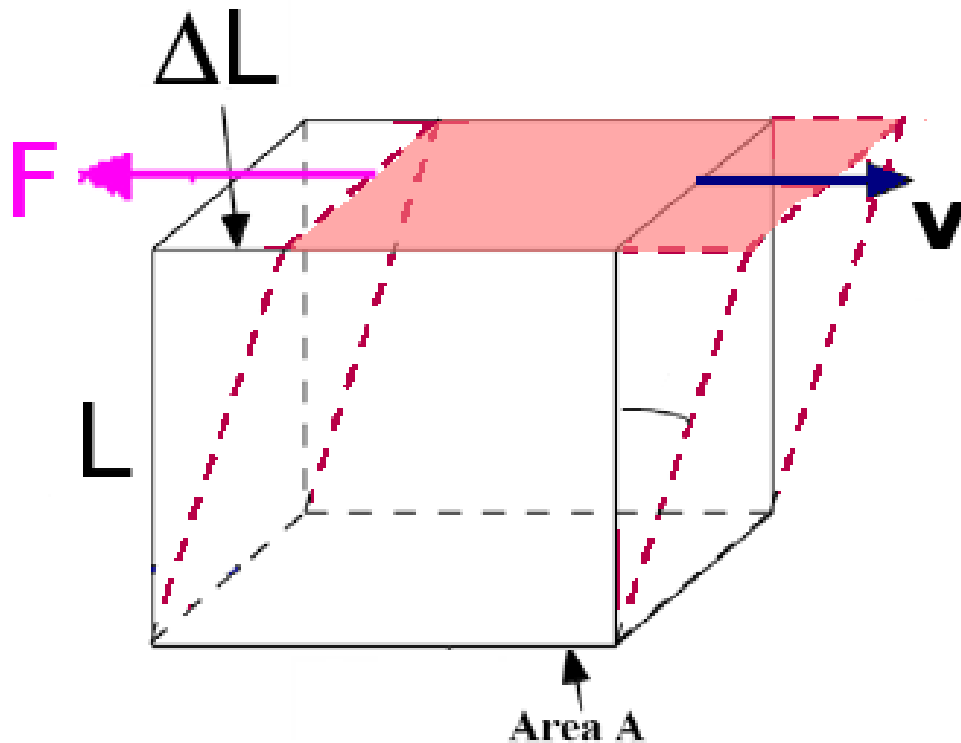


Phosphatidylcholine (lecithin)

Kilde: chestofbooks.com/food/science 200X

Viskositet

$$F = \text{viskositet} \frac{v A}{L}$$



Væske	Viskositet mPa s
Helium	0.0033 (4K)
Luft	0.0179 (15deg)
Vand	1.00 (20deg)
Honning	10000 (20deg)
Glycerin	1420 (20deg)
Glas (smelte)	10^6 (700deg)
Tjære	30×10^6 (20deg)
Glas	$10^{21} - 10^{24}$

Seje væsker

Tjære dråbe
eksperimentet.
Queensland universitet.

9 dråber er faldet siden
det startede i 1930.

Tjære er en blanding af
mere end 200 typer af
lange polymere.

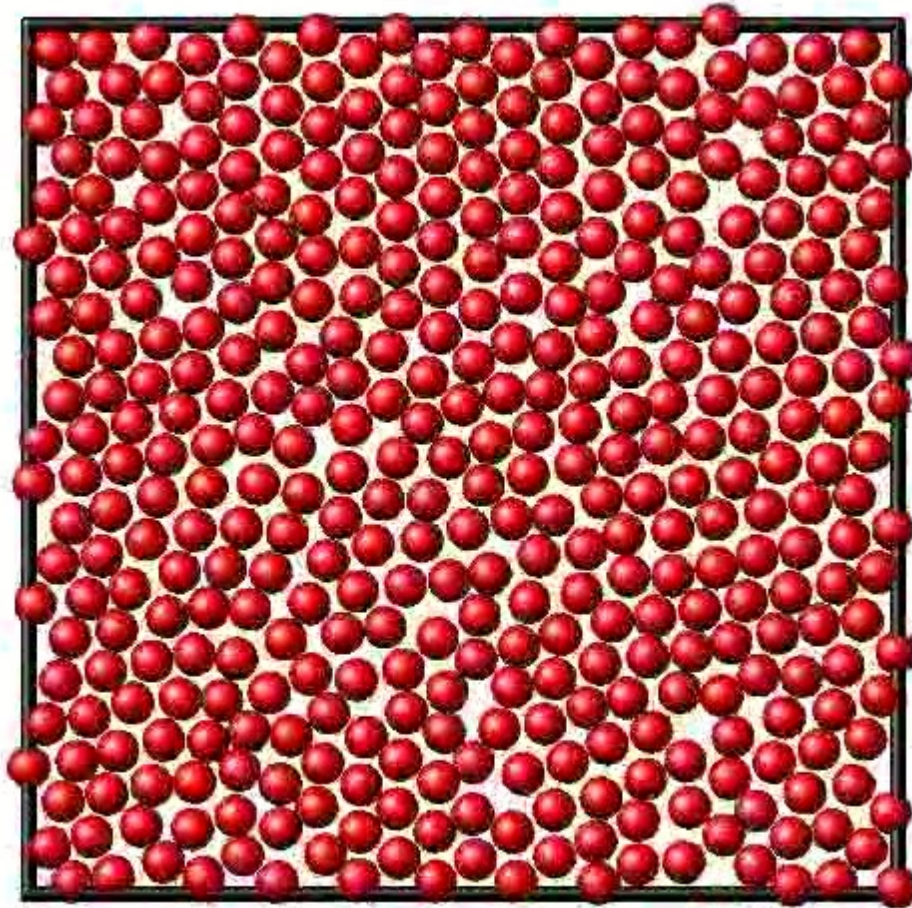


Time lapse film over et 1år.

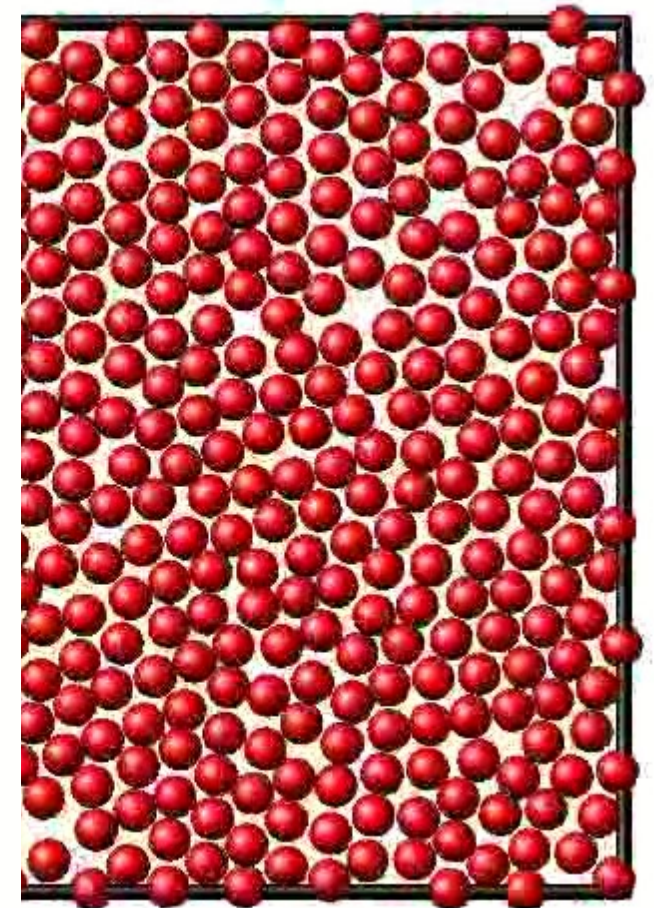


<http://smp.uq.edu.au/content/pitch-drop-experiment>

Krystal / glas



Langsom nedkøling



Ultrahurtig nedkøling

Glas har strukturen fra væsken, men er fråset. Ingen dynamik.

Fast OG Flydende



Opskrift:

Vand
Kartoffelmel

Kilde youtube: A pool filled with non-newtonian fluid

Fast OG flydende

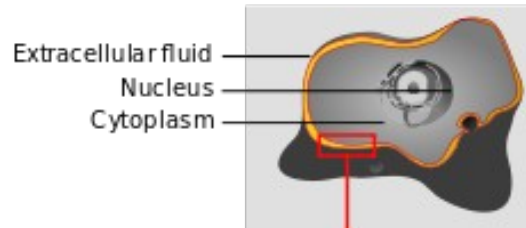


Fast OG flydende

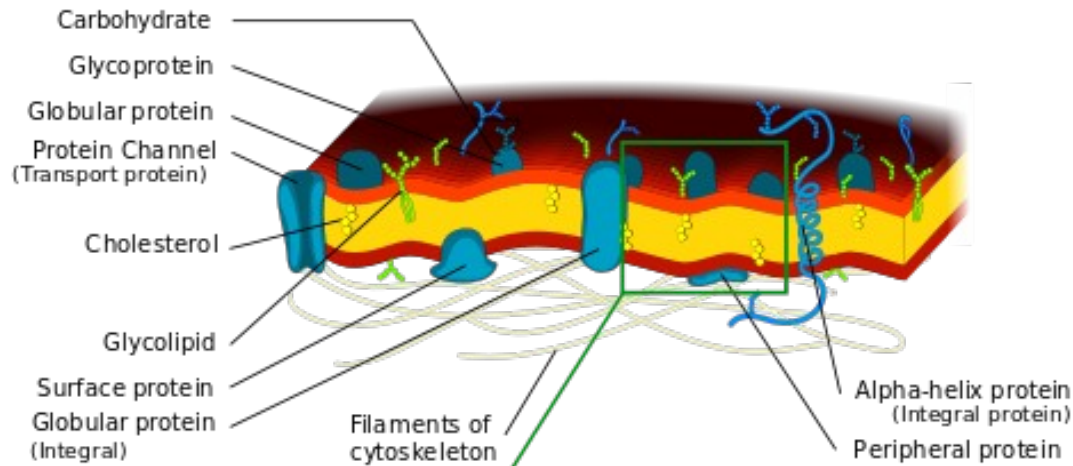


Biologiske Materialer

Cell



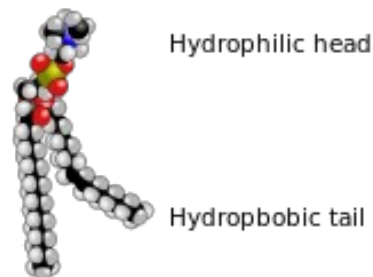
Cell membrane



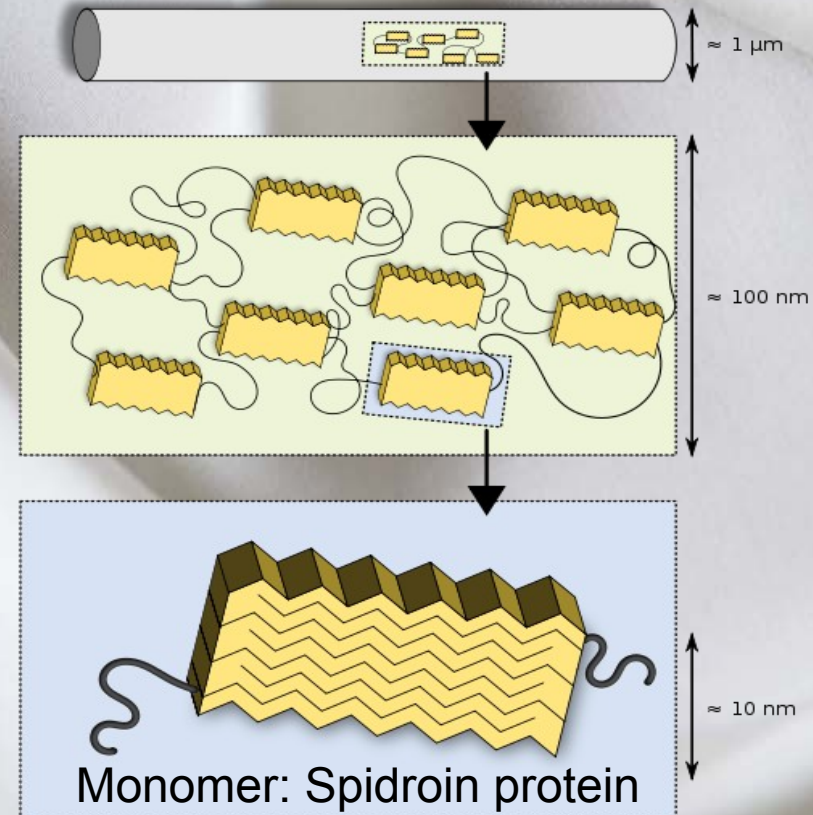
Phospholipid bilayer



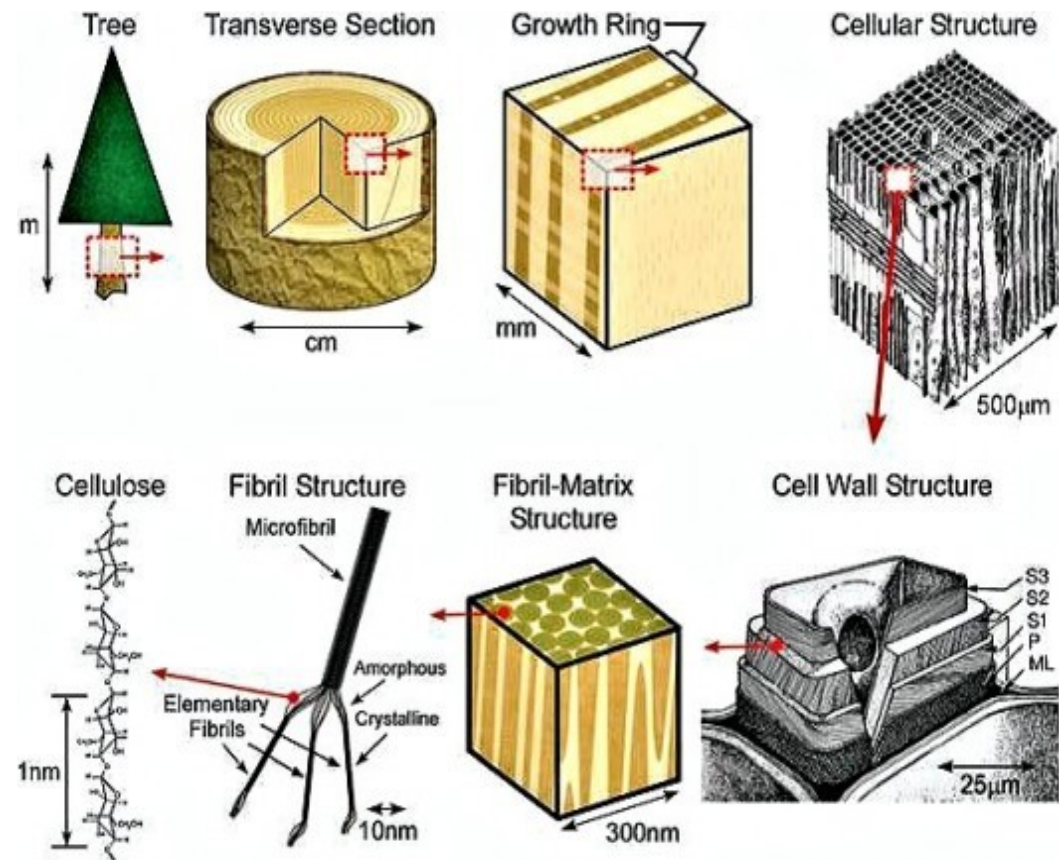
Phospholipid
(Phosphatidylcholine)



Naturlige polymere: Silke

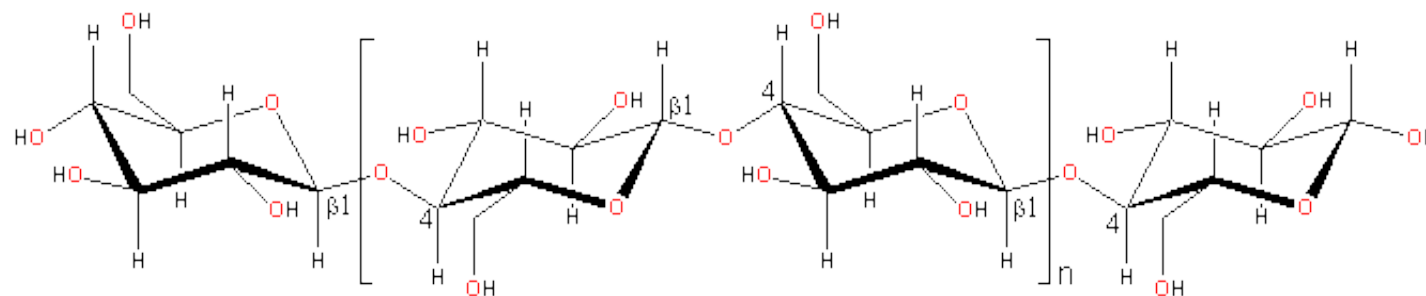


Naturlige polymere: Cellulose

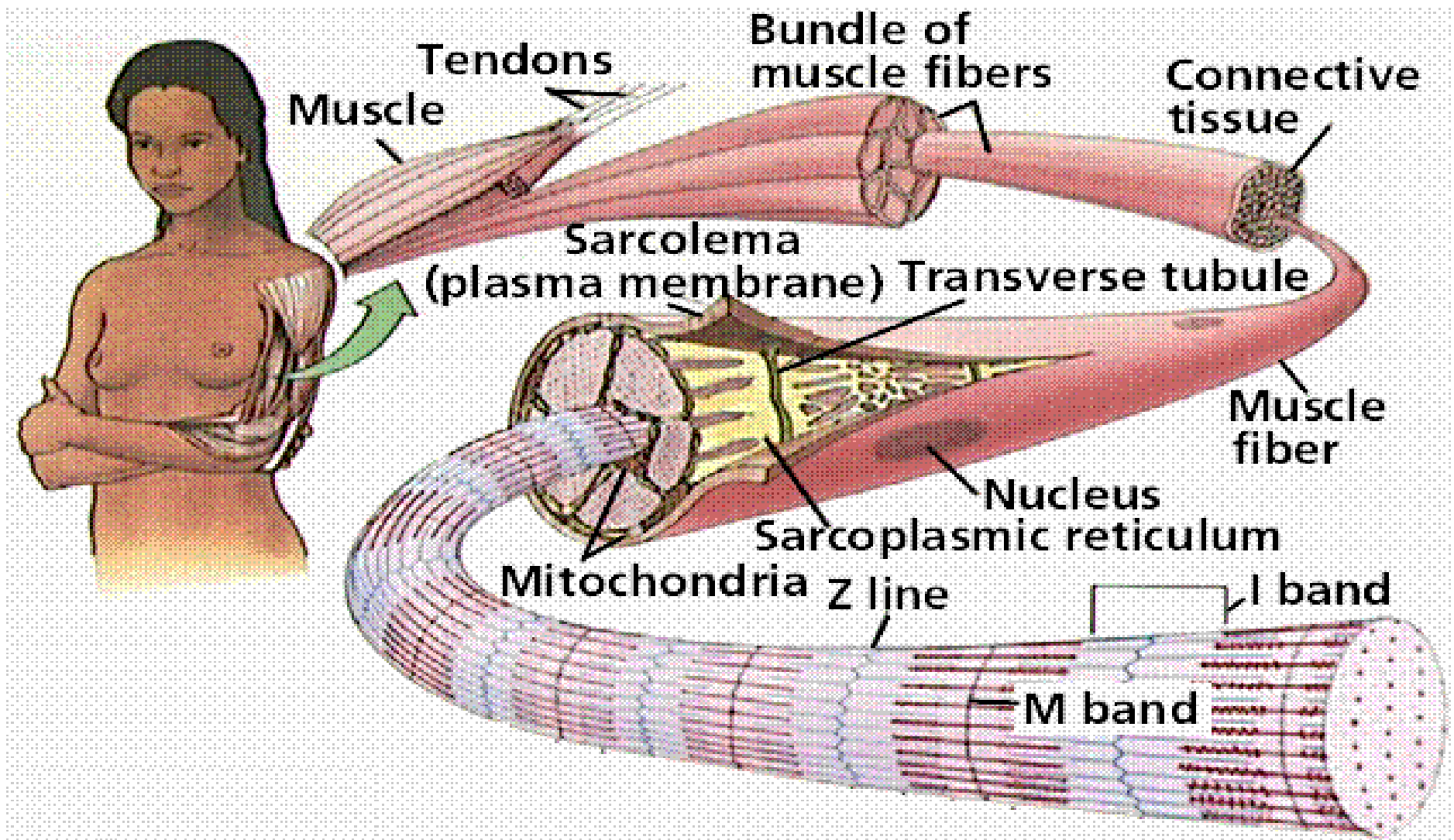


50% af vægten af træ er cellulose

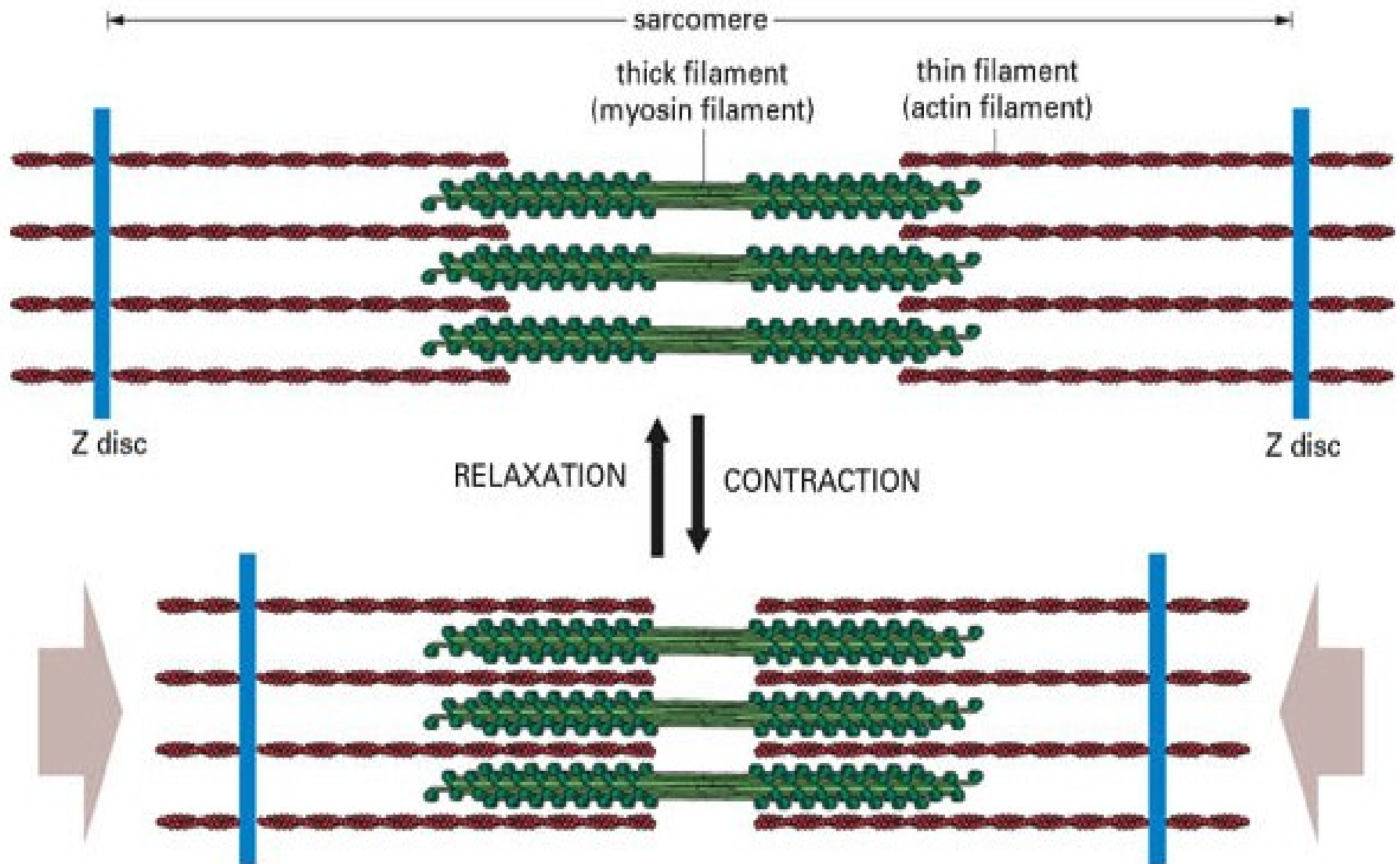
90% af vægten af bomuld er cellulose.



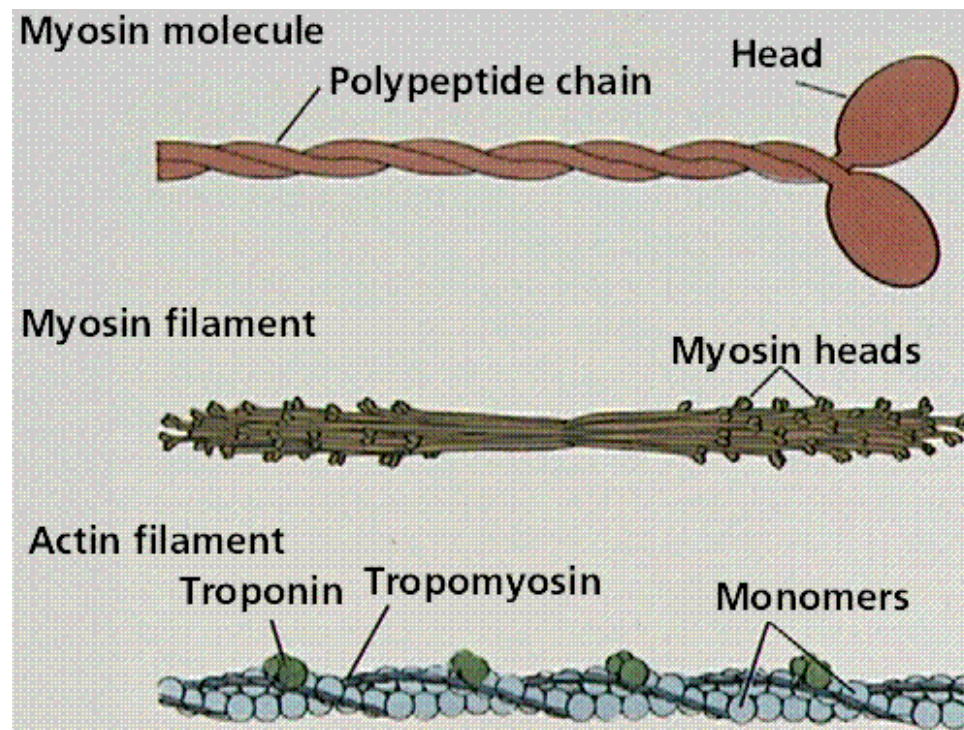
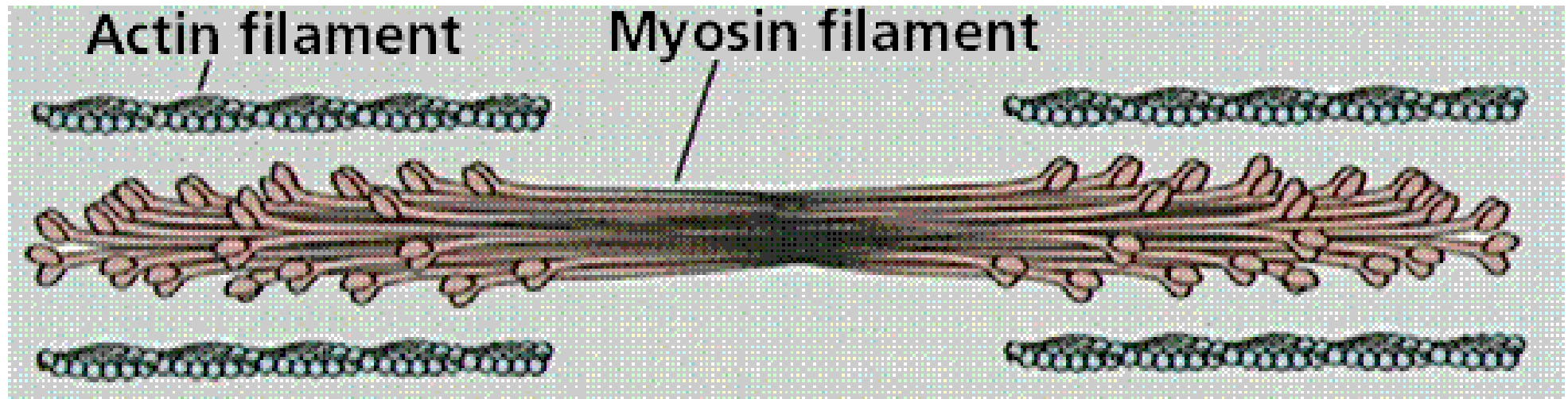
Muskler



Muskler



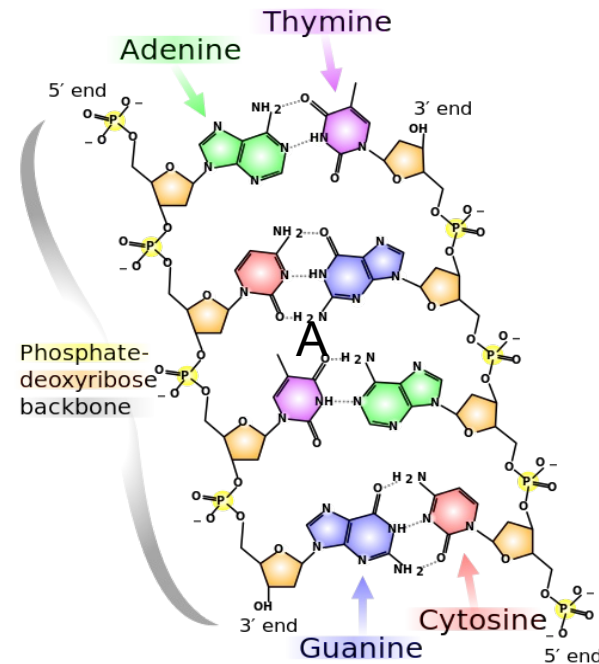
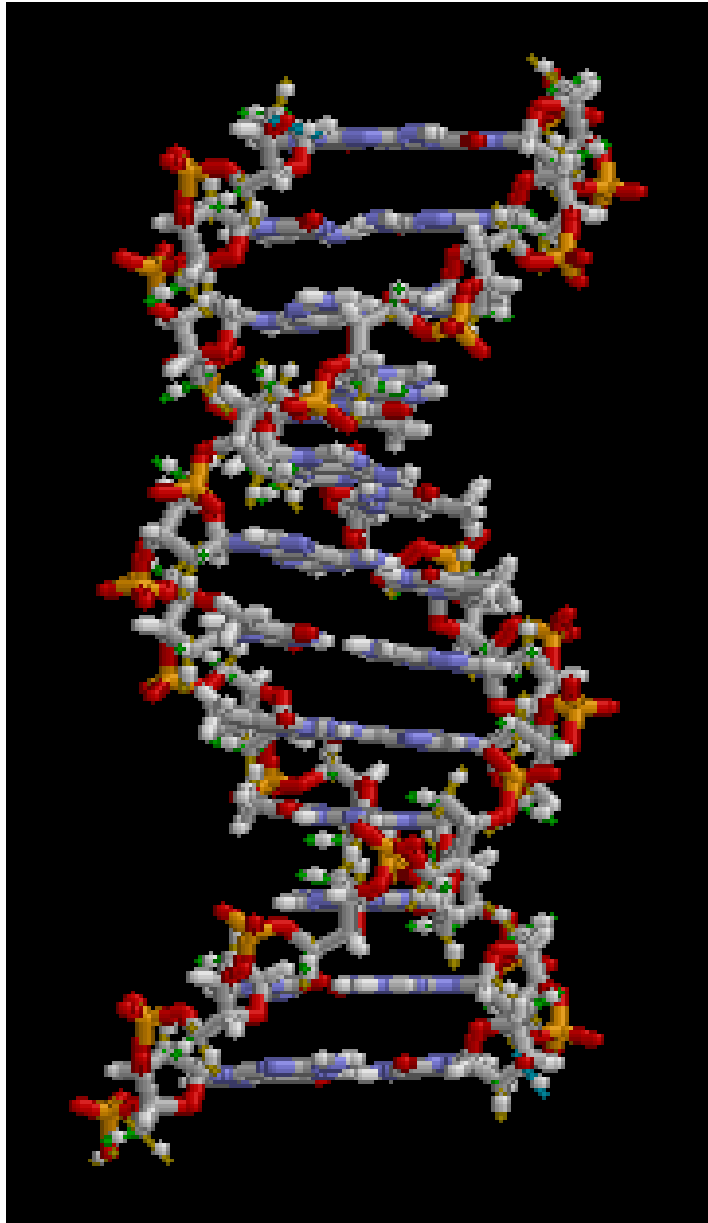
Muskel



Muskel

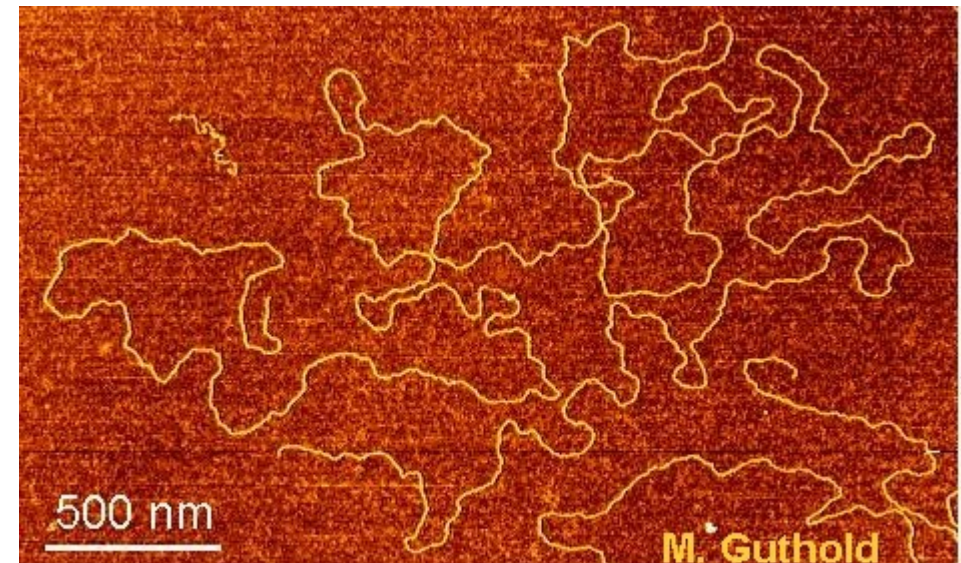


DNA



Polymer bestående af adenin, thymin, cytosin, guanin.

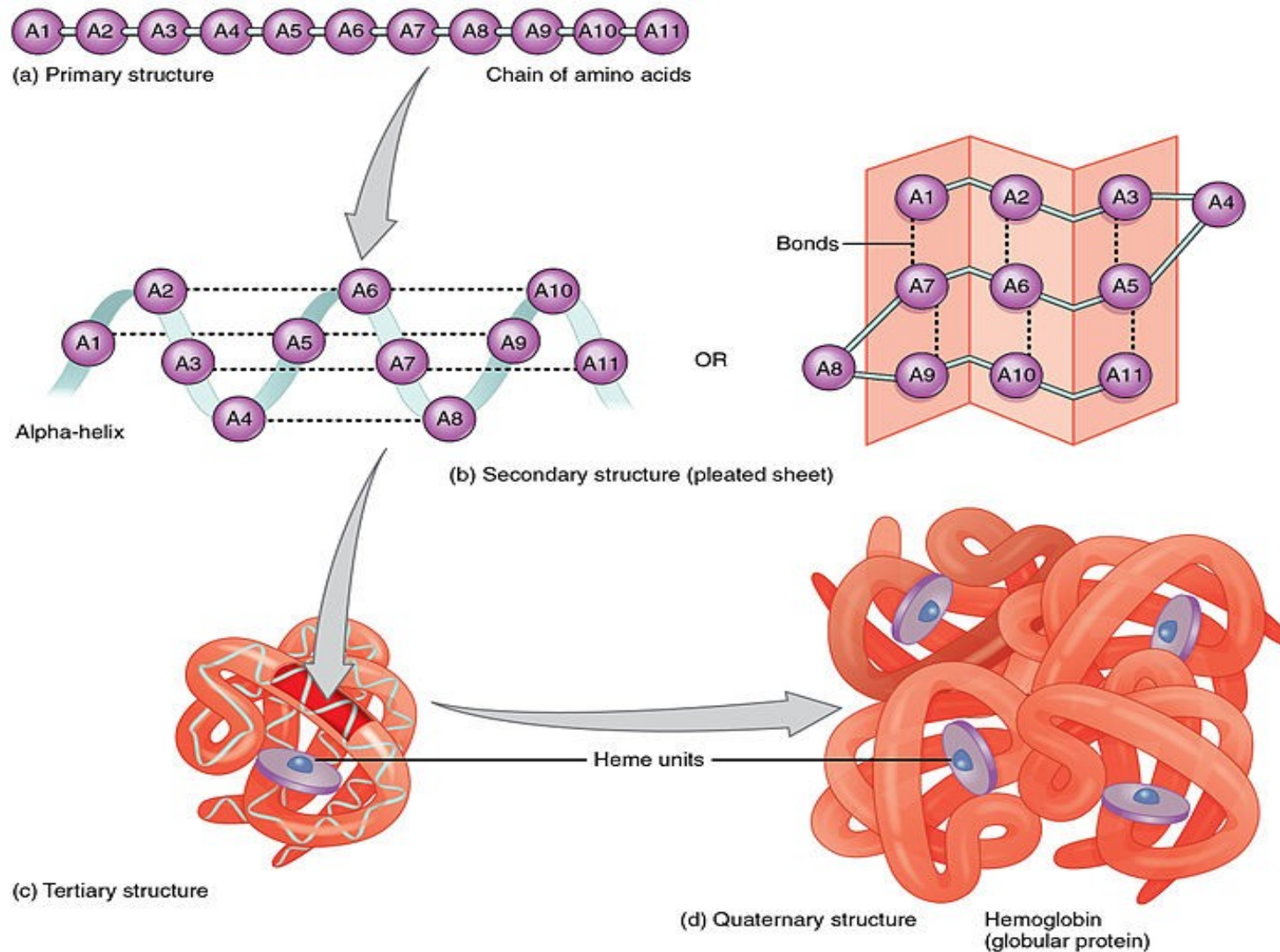
Menneske:
ca. 5 cm DNA i hvert kromosom x 46 kromosomer = 2 m.



AFM billed 16μm langt DNA

λ-DNA

Proteiner

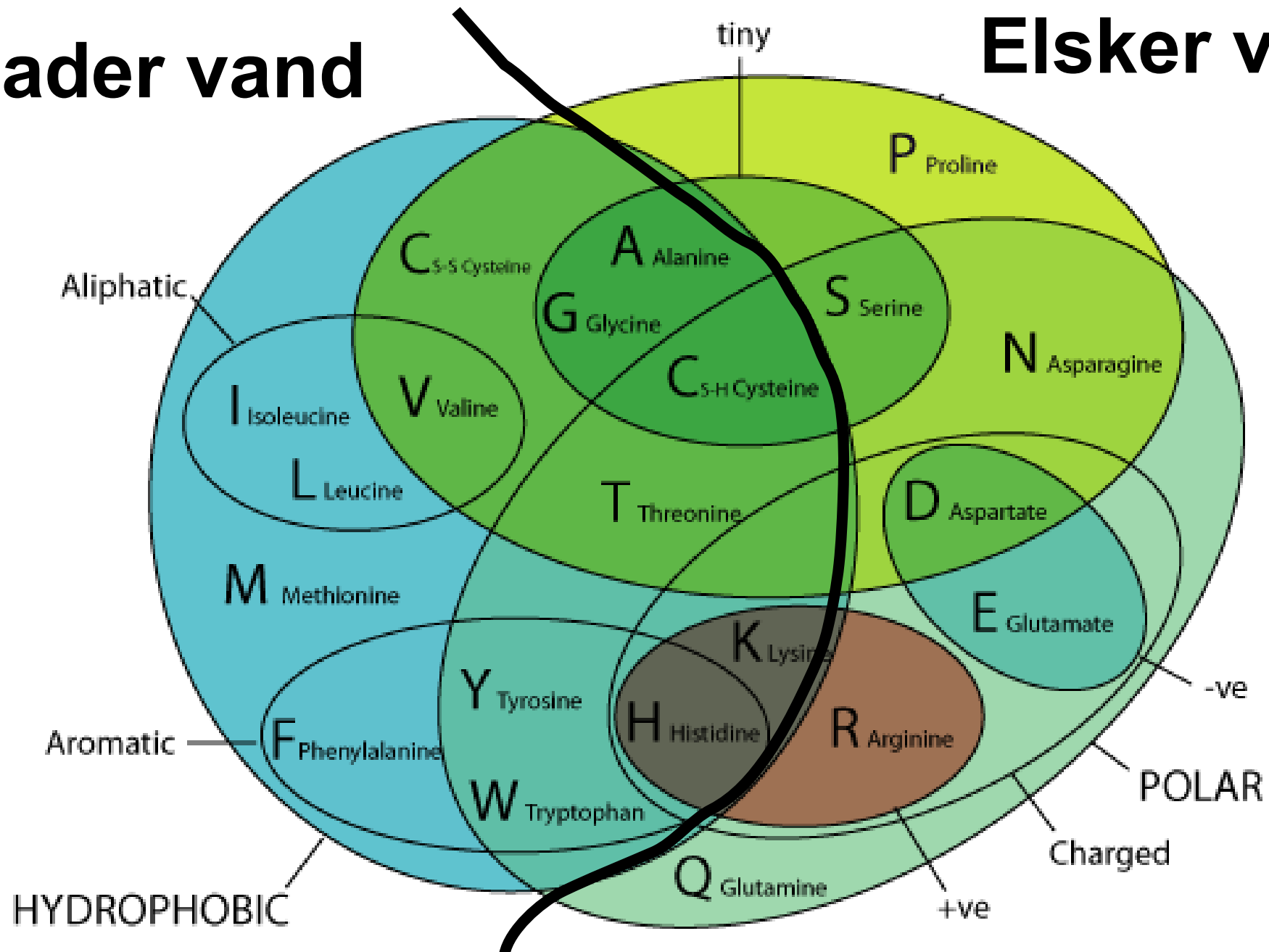


Sekvens af aminosyrer (20) bundet sammen med peptid bindinger.

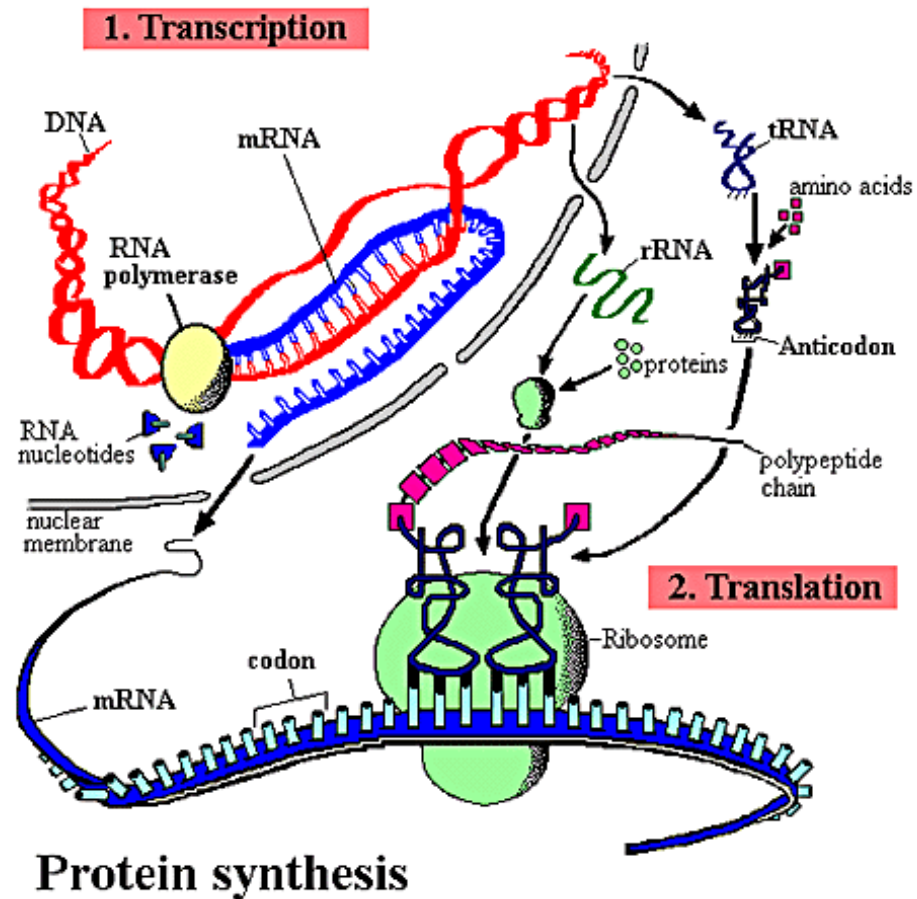
Aminosyrers forhold til vand

Hader vand

Elsker vand



Biologiens Central Dogma



DNA:

Information i base sekvens
ACATACACAACAACCACA



biologi

Protein

Aminosyre sekvens
MGHFTTEEDKATITSLWGKVN



selv-bygning

Foldning af protein med
specifik 3D struktur.

“Livets lego” klodser bygges
vha. selv-bygning!

Spørgsmål??